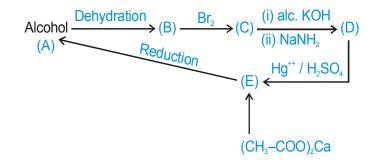


# SOLVED EXAMPLES

1. Find out unknown in following reactions.



Sol. Since E is obtained on dry distillation of calcium salt of acetic acid hence E will be CH<sub>3</sub>-C-CH<sub>3</sub>. Thus other unknowns are

$$H$$

$$A = CH_{3} - C - CH_{3}$$

$$OH$$

$$B = CH_{3} - CH = CH_{2}$$

$$C = CH_{3} - CH - CH_{2}$$

$$D = CH_{3} - C \equiv C - H$$

$$Br$$

$$Br$$

- 2. What will be structure of aromatic  $C_8H_8Cl_2$  (A), which on aqueous alkalihydrolysis gives product (B). (B) gives positive iodoform test.
- Sol. Since (B) is showing iodoform test hence it will be methylketone only as it is obtained on aqueous alkali hydrolysis of (A) which will be non-terminal gem dihalides as –

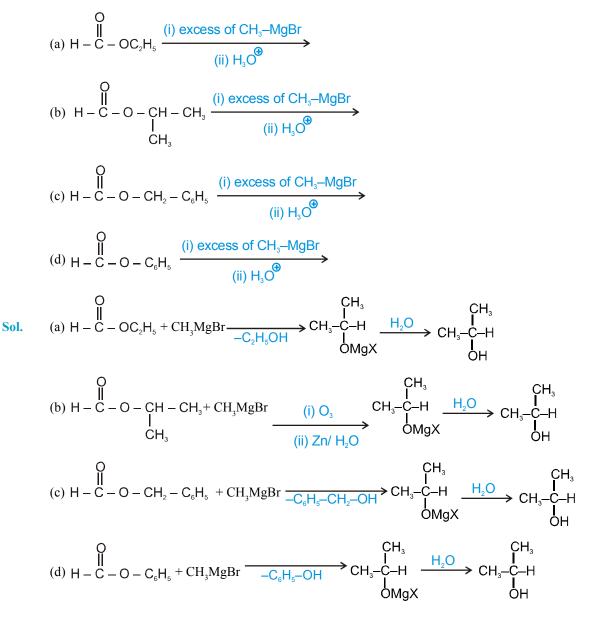
$$\begin{array}{c} \mathsf{R} - \mathsf{C} \mathsf{I} \\ \mathsf{I} \\ \mathsf{C} \mathsf{I} \\ \mathsf{C} \mathsf{I} \\ \mathsf{C} \mathsf{I} \\ \mathsf{C} \mathsf{I} \end{array} \xrightarrow{\mathsf{KOH}} \mathsf{R} - \mathsf{C} \mathsf{C} - \mathsf{CH}_{3} \xrightarrow{\mathsf{KOH}} \mathsf{R} - \mathsf{C} - \mathsf{ONa} + \mathsf{CHI}_{3} \xrightarrow{\mathsf{C}} \mathsf{O} \\ \overset{\mathsf{I}}{\mathsf{O}} \overset{\mathsf{O}}{\mathsf{O}} \xrightarrow{\mathsf{I}} \mathsf{R} - \mathsf{C} - \mathsf{ONa} + \mathsf{CHI}_{3} \xrightarrow{\mathsf{C}} \mathsf{O} \xrightarrow{\mathsf{I}} \mathsf{R} \xrightarrow{\mathsf{O}} \mathsf{O} \xrightarrow{\mathsf{I}} \mathsf{I} \xrightarrow{\mathsf{O}} \mathsf{O} \xrightarrow{\mathsf{I}} \mathsf{I} \xrightarrow{\mathsf{O}} \mathsf{O} \xrightarrow{\mathsf{I}} \mathsf{I} \xrightarrow{\mathsf{O}} \mathsf{O} \xrightarrow{\mathsf{I}} \mathsf{I} \xrightarrow{\mathsf{O}} \xrightarrow{\mathsf{O}} \mathsf{I} \xrightarrow{\mathsf{O}} \xrightarrow{\mathsf{O}} \xrightarrow{\mathsf{O}} \mathsf{I} \xrightarrow{\mathsf{O}} \xrightarrow{\mathsf{$$

Now unknown 'R' can be known as :

$$R - C_{B}^{I} - C_{B}^{I} = C_{B}^{I} H_{B}^{I} Cl_{2}$$

$$R = C_{B}^{I} H_{B}^{I} Cl_{2} - C_{2}^{I} H_{3}^{I} Cl_{2} = C_{B}^{I} H_{5}$$
Hence 'A' is  $C_{B}^{I} H_{5} - C_{B}^{I} - C_{B}^{I}$ 

3. Write the products of the following reactions.



**4**. What will be hydration and hydroboration product for Cyclohexylethyne.

$$\bigcup_{Cyclohexylethyne} C C - H \xrightarrow{Hg^{2+} / H_2SO_4} \bigcup_{Cyclohexylmethylketone} C - CH_3$$

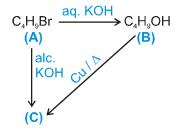
Sol. Cyclohexylethyne

(i) 
$$B_2H_6$$
 or  $Sia_2BH$   
(ii)  $H_2O_2$  (OH<sup>-</sup>)  
Cyclobexylethanal

- 5. Which hydrocarbon on ozonolysis gives acetone only ?
- **Sol.** Acetone only, means two moles of acetone.

$$CH_{3} - C = C - CH_{3} \xrightarrow{1. \text{ PCI}_{6}/\text{Ether}} A \xrightarrow{H_{2}O} B + C CH_{3} - C = O + O = C - CH_{3}$$
$$\underset{CH_{3}}{\overset{I}{\longrightarrow}} H_{2}O \xrightarrow{I} B + C CH_{3} - C = O + O = C - CH_{3}$$

6. Predict the structure of (A) in the following sequence :



- **Sol.** Since (B) is alcohol and (C) is alkene hence (B) is  $3^{\circ}$  alcohol only according to question (It is known that alkene can only be obtained from  $3^{\circ}$  alcohol when heated with copper). Thus structure of (B) is  $(CH_3)_3C$ -OH and its corresponding. alkyl bromide will be  $(CH_3)_3C$ -Br (tertiarybutylbromide)
- 7. What will be structure of  $C_4H_8O_2$  which on treating with excess  $CH_3$ -MgBr followed by acidification gives sole alcohol (A). (A) on treating with sodium hypoiodite solution gives positive iodoform test.
- Sol. Since (A) gives positively iodoform test hence it will be alkanol-2. 2° alcohol can be obtained only when alkylformate is treated with Grignard's reagent via aldehyde where alkyl part is alkyl part of Grignard's reagent. As Grignard's reagent is  $CH_3$ -MgBr hence 2° alcohol will be  $CH_3$ -CH-OH (propanol-2). Thus  $C_4H_8O_2$  is either  $CH_3$ -CH-OH (propanol-2).

$$\begin{array}{c} O & CH_3 \\ \parallel & \parallel \\ H-C-OCH_2-CH_2-CH_3 & \text{or} & H-C-O-CH-CH_3 \end{array}$$
Reactions:

Here we get two alcohols propanol-2 and propanol-1. Alkyl part of formic acid ester which gives propanol-2 will

be isopropyl only. Thus structure of  $C_4H_8O_2$  is  $H - C - OCH (CH_3)_2$ 

$$Ph \qquad C = N \qquad \xrightarrow{OH} \qquad \underbrace{1. \operatorname{PCl}_{6}/\operatorname{Ether}}_{2. \operatorname{H}_{2}O} A \xrightarrow{\operatorname{H}_{3}O^{+}} B + C$$

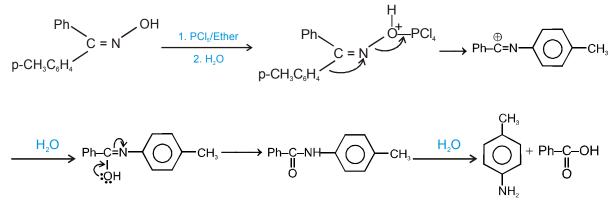
$$P-CH_{3}C_{6}H_{4} \qquad \underbrace{2. \operatorname{H}_{2}O} A \xrightarrow{\operatorname{H}_{3}O^{+}} B + C$$

$$A, B, C \text{ are}$$

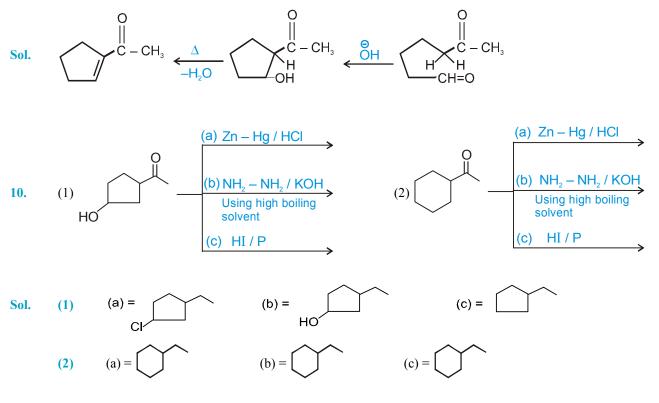
$$(A) \operatorname{PhCONH-p} - \operatorname{CH}_{3}C_{6}H_{4} \qquad (B) \operatorname{PhCOOH}$$

$$(C) \operatorname{pCH}_{3}C_{6}H_{4}\operatorname{NH}_{2} \qquad (D) \operatorname{PhCHO}$$

Sol. (A,B,C)



9. Which carbonyl compound on heated with dilute alkali gives 1- acetylcyclopentene.



- 11. (A) on treating with (B) in the presence of dry ether gives (C) which on acids hydrolysis gives (D). (D) on oxidation gives 2,5-dimethylhexan-3-one.
- Sol.

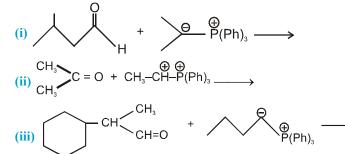
By knowing structure of given product (D) will be  $CH_3 - CH - CH_2 - CH - CH - CH_3$  hence (C) will be  $\begin{vmatrix} I \\ I \\ CH_3 \end{vmatrix}$  OH  $CH_3$ 

 $CH_3 - CH - CH_2 - CH - CH_3$  and finally A & B will have following two structures.  $| I - CH_3 = 0$   $CH_3 = 0$  $CH_3 = 0$ 

$$A = CH_3 - CH - CH_2 - CH = O \quad or \quad A = CH_3 - CH - CH = O$$

$$\begin{array}{ccc} B = \mathsf{CH}_3 - \mathsf{CH} - \mathsf{MgBr} & \text{or} & B = \mathsf{CH}_3 - \mathsf{CH} - \mathsf{CH}_2 - \mathsf{MgB} \\ & & & & \\ \mathsf{CH}_3 & & & \mathsf{CH}_3 \end{array}$$

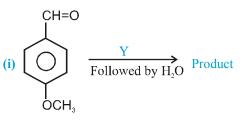
**12.** Predict the product for the followings :

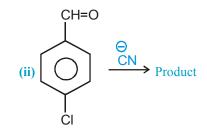


**Sol.** Witting reaction

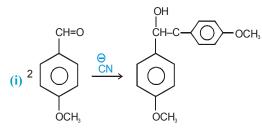


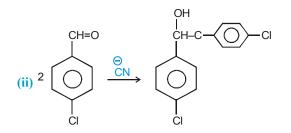
**13.** Predict Product –





Sol. Benzoin condensation reaction



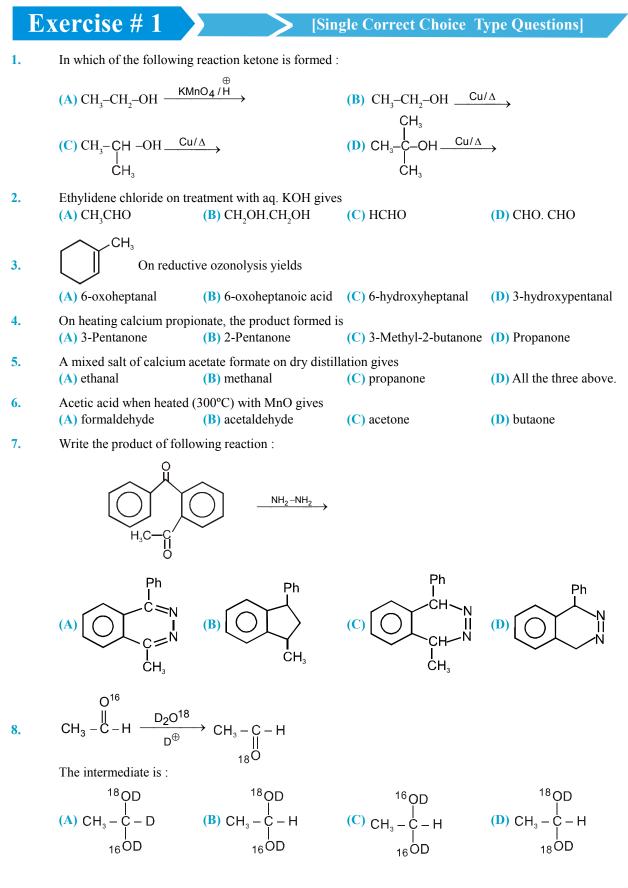


**14.** Predict product for the following

$$C_{6}H_{5} C = O \xrightarrow[C_{6}H_{5} - C - O - H]{H_{5} - C - O - H} Product$$

**Sol.** Baeyer-villiger oxidation

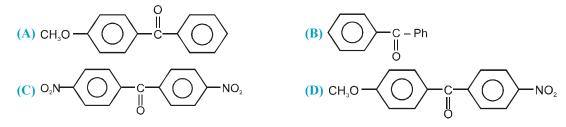
$$CH_{3}-C-OC_{6}H_{5} + C_{6}H_{5} - C - OH$$



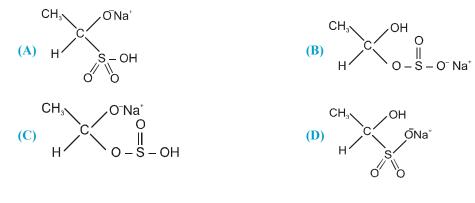
- 9. Aromatic carbonyl compounds having molecular formula C<sub>8</sub>H<sub>8</sub>O react with NH<sub>2</sub>OH how many oximes can be formed :
  - (A) 8 (B) 10 (C) 12 (D) 6
- 10.Arrange the following compounds in decreasing orders of rate of exchange of  $O^{18}$  with  $H_2O^{18}$ (X) CCl\_3CHO(Y) CH\_3CHO(Z) CH\_3COCH\_3(W) CF\_3CHO

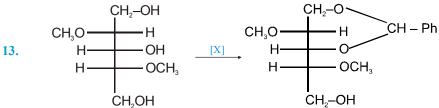
(A) W > Z > X > Y (B) W > X > Y > Z (C) W > Y > Z > X (D) W > Z > Y > X

11. Which of the following compound has the largest equilibrium constant for the addition of water ?



12. Acetaldehyde on reaction with sodium hydrogen sulphite produces



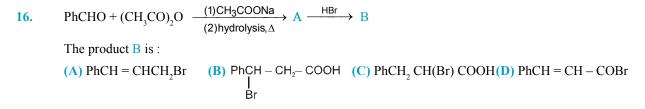


Compound (X) in the above reaction.

14.

The correct order of reactivity of PhMgBr with following compounds will be.							
$(C_6H_5)_2CO,$	CH	$CH_3 - CH = O$ ,		$(CH_3)_2 C = O$			
(1)	(2)		(3)				
(A) $1 > 2 > 3$	<b>(B)</b> $2 > 3 > 1$	(C) $3 > 2 > 1$		<b>(D)</b> $1 > 3 > 2$			

15.The cyanohydrin of a carbonyl compound on hydrolysis gives lactic acid. The carbonyl compound is(A) HCHO(B) CH<sub>3</sub>CHO(C) CH<sub>3</sub>COCH<sub>3</sub>(D) CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>



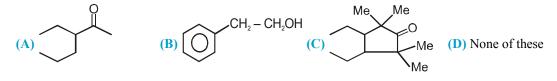
17. In the given reaction the product is :

$$CH_{3}-CH_{2}-C-OCH_{3} + CH_{3}-C-OCH_{3} \frac{(1) CH_{3}O^{\Theta} Na^{*}}{(2) CH_{3}OH} \xrightarrow{H_{2}O}{\Delta}$$

$$(A) CH_{3}-CH_{2}-C-CH_{2}-C-OCH_{3} \qquad (B) CH_{3}-C-CH-COOCH_{3}$$

$$(C) H-C-CH_{2}-CH-COOCH_{3} \qquad (D) CH_{3}-C-CH_{2}-CH_{3}$$

**18.** The compounds that undergo Aldol condensation is :

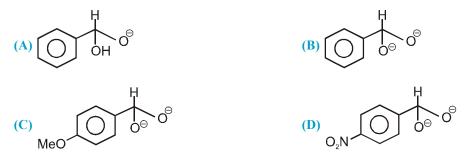


- 19.Which of the following will not undergo aldol condensation ?(A) CH<sub>3</sub>CHO(B) CH<sub>3</sub>CH<sub>2</sub>CHO(C) CD<sub>3</sub>CHO(D) PhCHO
- 20. (X) is the product of cross aldol condensation between benzaldehyde ( $C_6H_5CHO$ ) and acetone What is its structure ?

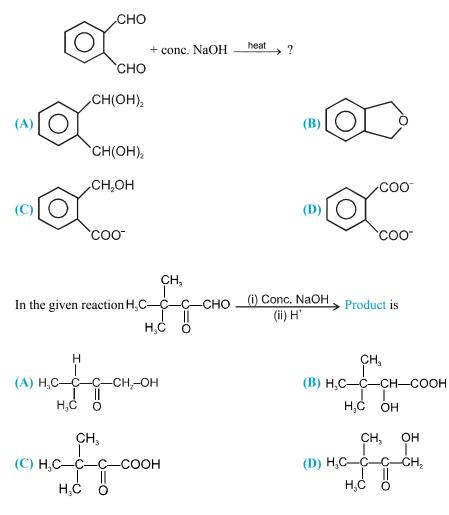
$$\begin{array}{c} O \\ \parallel \\ (A) C_6H_5-CH=CH-C-CH_3 \\ (C) C_6H_5-CO-CH_2-C=(CH_3)_2 \\ \end{array}$$

$$\begin{array}{c} O \\ (B) C_6H_5-CH=C-(CH_3)_2 \\ (D) \text{ None of these} \\ \end{array}$$

- In which of the following compounds the methylene hydrogens are the most acidic ?
   (A) CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>
   (B) CH<sub>3</sub>CH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub>
   (C) CH<sub>3</sub>CH<sub>2</sub>CH(COOC<sub>2</sub>H<sub>5</sub>)
- (D) CH<sub>3</sub>COCH<sub>2</sub>CN.
- 22. In the cannizzaro's reaction the intermediate that will be the best hydride donor ?

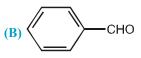


23. Product of following reaction is



25. Cannizzaro reaction does not take place with

(A) (CH<sub>3</sub>)<sub>3</sub>CCHO.

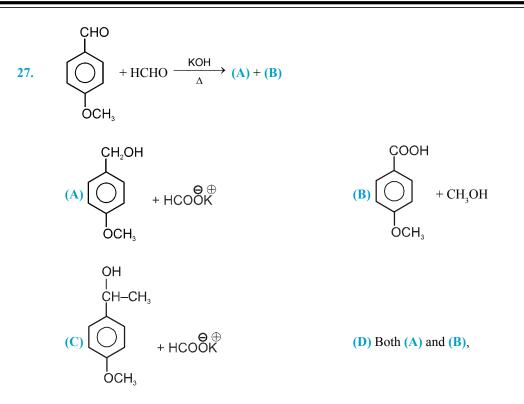


С)

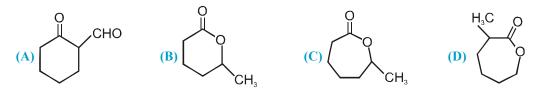
26. In the reaction,  $(CH_3)_3CCHO + HCHO \xrightarrow{\text{NaOH}} A + B$ .

the products (A) and (B) are respectively :

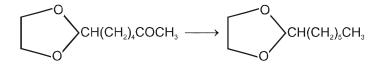
- (A)  $(CH_3)_3CCH_2OH$  and  $HCOO^-$  Na<sup>+</sup>.
- (B) (CH<sub>3</sub>)<sub>3</sub>CCOONa and CH<sub>3</sub>OH.
- (C) (CH<sub>3</sub>)<sub>3</sub>CCH<sub>2</sub>OH and CH<sub>3</sub>OH.
- **(D)**  $(CH_3)_3$ COONa and HCOO<sup>-</sup> Na<sup>+</sup>.



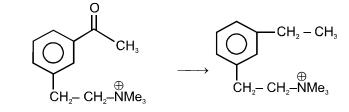
28. 2-Methyhlcyclohexanone is allowed to react with metachloroperbenzoic acid. The major product in the reaction is



**29.** In the following conversion

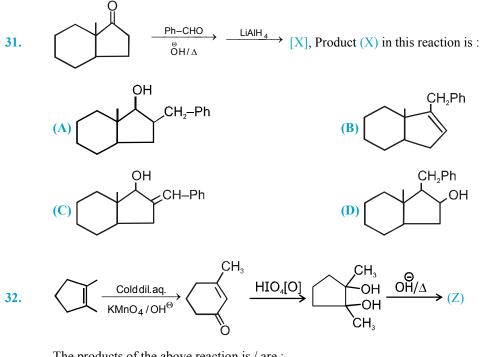


Which of the following regents is suitable ?(A) NH2NH2,KOH, DMSO(B) NaBH4(C) Zn-Hg, concentrated H2SO4(D) LiAlH4

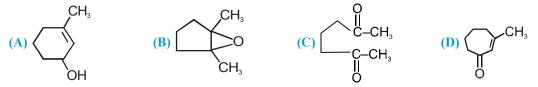


Above conversion can be achieved by (A) NH<sub>2</sub>-NH<sub>2</sub>/ NaOH (B) Zn-Hg/HCl (C) LiAlH<sub>4</sub>

**(D)** NaBH<sub>4</sub>.



The products of the above reaction is / are :

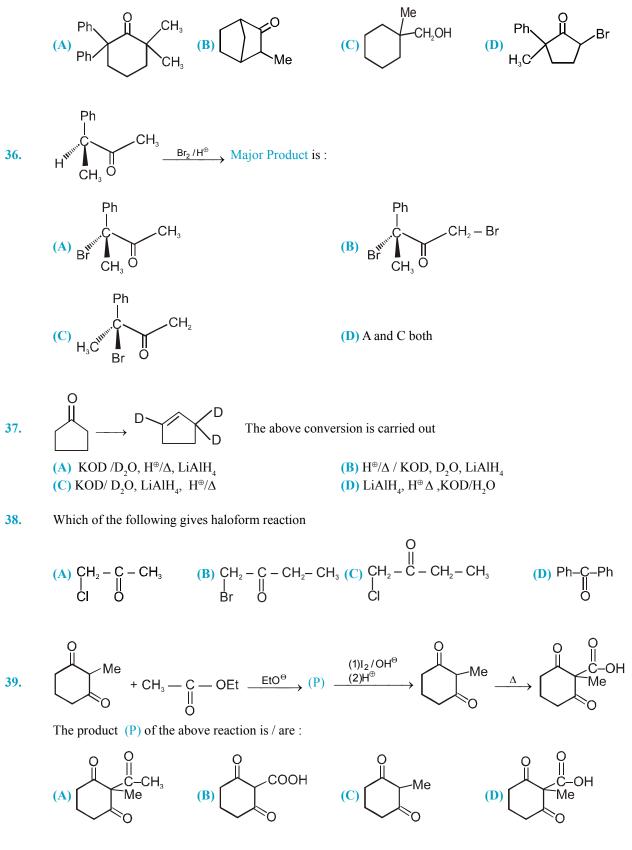


33. What will be the product of the following reaction

(A) Ph-CH-O-C-Me (B) Ph C -CH 
$$\stackrel{\text{RCO}_{3}H}{\underset{\text{Me}}{\overset{\text{RCO}_{3}H}{\overset{\text{RCO}_{$$

In which of the following reaction deuterium exchange is observed ? 34.

(A) 
$$CH_{3} - C - CH_{2} - CH_{3} \xrightarrow{H_{1}O/OH^{\Theta}} CD_{3} - C - CD_{2} - CH_{3}$$
  
(B)  $CH_{3} - CH - C - H \xrightarrow{D_{2}O/OH} CH_{3} - CD - C - H$   
Ph O  
(C)  $\xrightarrow{P_{1}} D_{2}O/OH \xrightarrow{P_{2}} D \xrightarrow{P_{1}} H$   
(D)  $H - C - H \xrightarrow{D_{2}O/OH} D - C - D$ 



In which of the following reaction deuterium exchange is not observed ?

40. 2-pentanone can be distinguished from 3- pentanone by the reagent ?
(A) 2, 4- Dinitrophenyl hydrazine
(B)Tollen's reagent
(C) I<sub>2</sub> and dilute NaOH
(D) NaHSO<sub>3</sub>

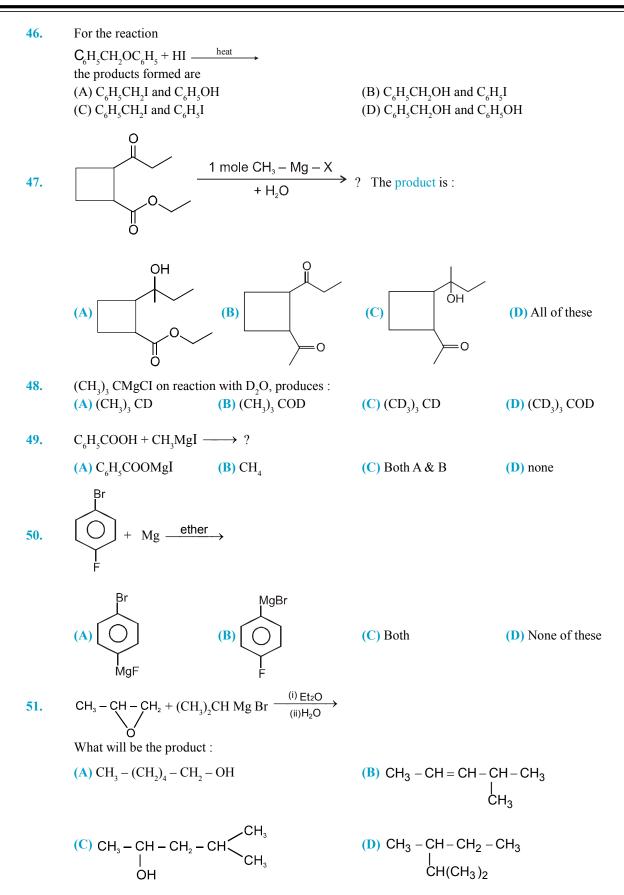
41. 
$$\begin{array}{c} & \bigoplus_{i=1}^{Q} + (C_{6}H_{5})_{3}P = CHCH_{3} \longrightarrow \\ & \bigoplus_{i=1}^{Q} - CH_{3} & \bigoplus_{i=$$

$$C_{6}H_{5}CHO + CH_{3}NO_{2} \xrightarrow{\text{NaOH}} (X)$$
(A)  $C_{6}H_{5} - CH - CH - NO_{2}$ 
(B)  $C_{6}H_{5}CHCH_{2}NO_{2}$ 
(C)  $C_{6}H_{5}CH = CH - NO_{2}$ 
(D)  $C_{6}H_{5}CH \xrightarrow{CH_{2}NO_{2}} CH_{2}NO_{2}$ .

**43.** What is the final product of this sequence of reactions ?

44. Ph-CH<sub>2</sub>-COOEt + 
$$\begin{array}{c} O\\ EtO\end{array}$$
 (B) Ph-CH<sub>2</sub>-COOEt (C) Ph-CH (D) None of these (D) None of these

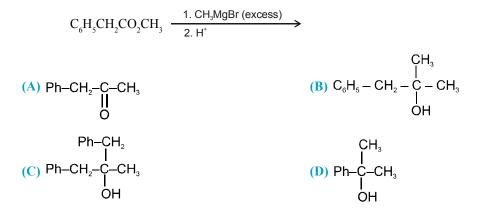
45. Which of the following will gives iodoform with NaOI ?



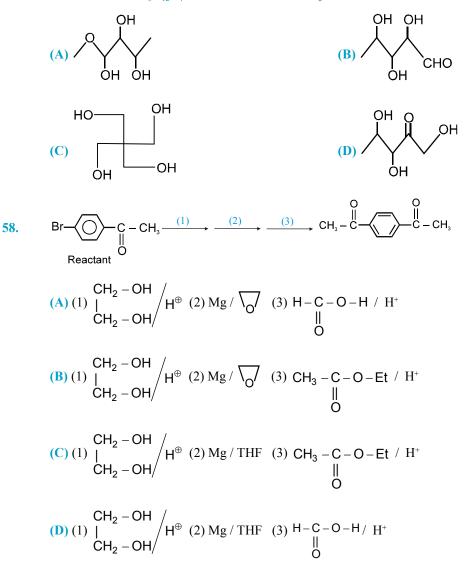
54. How many functional group produced  $CH_4$  gas by the reaction of compound (I) with  $CH_3MgBr$ .

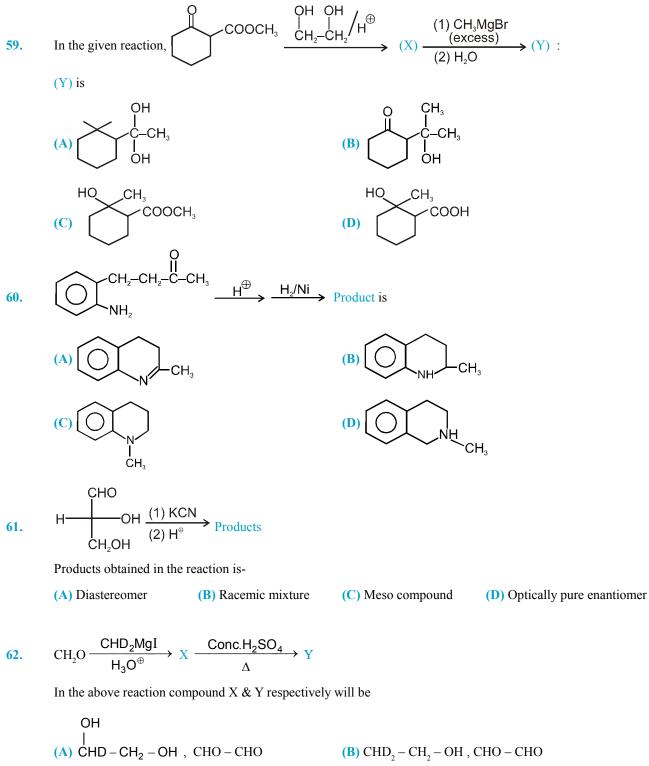
OH HO--CH₃ οŢ\_\_\_ SO<sup>®</sup>H HOOC-۰N **(I**) **(A)** 3 **(B)** 4 **(C)** 5 **(D)** 6  $\begin{array}{c} O \\ \parallel \\ Ph-MgBr + (CH_3)_2CH - C - CI \longrightarrow [X] \xrightarrow{Zn-Hg/HCI} [Y] \end{array}$ 1 mole Identify structure of [Y].  $\begin{array}{c} \textbf{(B)} \ \mathsf{Ph}-\mathsf{CH}-\mathsf{CH}_2-\mathsf{CH}_3\\ & \downarrow\\ \mathsf{CH}_3 \end{array}$ (A)  $Ph - CH_2 - CH - CH_3$ | CH3  $(C) Ph-C-CH_3 \\ | \\ CH_3 \\ CH_3$ (**D**)  $\bigcirc$   $-CH_2 - CH_2 - CH_3$  $CH_3$ 

56. Predict the major product in the following reaction:



57. A compound X  $(C_5H_{12}O_4)$  upon treatment with CH<sub>3</sub>MgX gives 4 mole of methane. Identify the structure of (X).



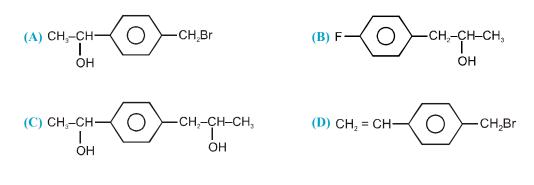


(C)  $CHD_2 - CH_2 - OH$ ,  $CD_2 = CH_2$ (D)  $CHD - CH_2 - OH$ ,  $CD_2 = CH_2$ 

ÒН

63. For the given reaction 
$$F \longrightarrow CH_2Br + Mg \xrightarrow{dry \text{ ether}} A \xrightarrow{(i)CH_3CHO}_{(ii)H_3O^+} B$$

product **B** is :

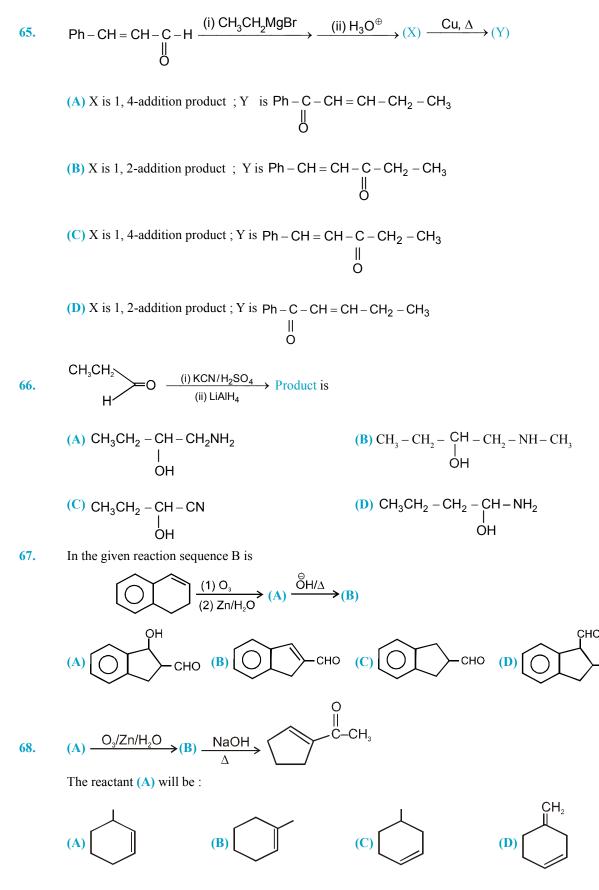


64. Consider the following sequence of reactions-.

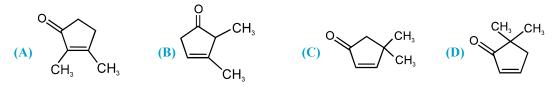
$$PhC \equiv CH \xrightarrow{HgSO_4} A \xrightarrow{NH_2OH} B + C.$$

The products (A), (B) and (C) are respectively,

(A) PhCHO,  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$  and  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$ (B) PhCH<sub>2</sub>CHO,  $\stackrel{PhCH_2}{\longrightarrow} C = N \quad OH$  and  $\stackrel{PhCH_2}{\longrightarrow} C = N \quad OH$ (C)  $\stackrel{Ph}{\longrightarrow} C = O$ ,  $\stackrel{Ph}{\longrightarrow} C \stackrel{OH}{\longrightarrow} OH$  and  $\stackrel{Ph}{\longrightarrow} C = N - OH$ (D)  $\stackrel{Ph}{\longrightarrow} C = O$ ,  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$  and  $\stackrel{Ph}{\longrightarrow} C = N - OH$ 

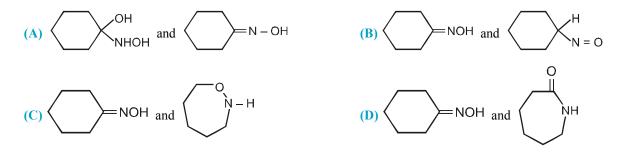


69. 
$$CH_3 \xrightarrow{\text{KOH, H}_2O} Product (C_7H_{10}O) :$$
  
 $CH_3 \xrightarrow{\text{CH}_3} CH_3 \xrightarrow{\text{KOH, H}_2O} Product (C_7H_{10}O) :$ 



70. Consider the following sequence of reactions :

The products (A) and (B) are, respectively :



71. Compound (X)  $C_9H_{10}O$  gives yellow coloured ppt with 2,4 DNP but does not give red coloured ppt with Fehling's solution. (X) on treatment with  $NH_2OH/H^+$  gives compound (Y)  $C_9H_{11}NO$ . (Y) when treated with  $PCl_5$  gives isomeric compound (Z). (Z) on hydrolysis gives propanoic acid and aniline. What will be the correct structure of (X), (Y) and (Z) ?

(A) 
$$C_{6}H_{5} - C - C_{2}H_{5};$$
  $C_{6}H_{5} - C - C_{2}H_{5};$   $C_{6}H_{5} - C - C_{2}H_{5};$   $C_{6}H_{5} - C - NH - C_{2}H_{5}$   
 $|| O OH - N O$ 

(C) 
$$C_6H_5 - CH_2 - C - CH_3$$
;  $C_6H_5 - CH_2 - C - CH_3$ ;  $CH_3 - C - CH_2 - NH - C_6H_5$   
 $\| \\ O \\ N - OH \\ O \\ O$ 

(D) 
$$C_6H_5 - C - C_2H_5$$
;  $C_6H_5 - C - C_2H_5$ ;  $C_6H_5 - C - NH - C_2H_5$   
 $|| \\ O \\ N - OH \\ || \\ O$ 

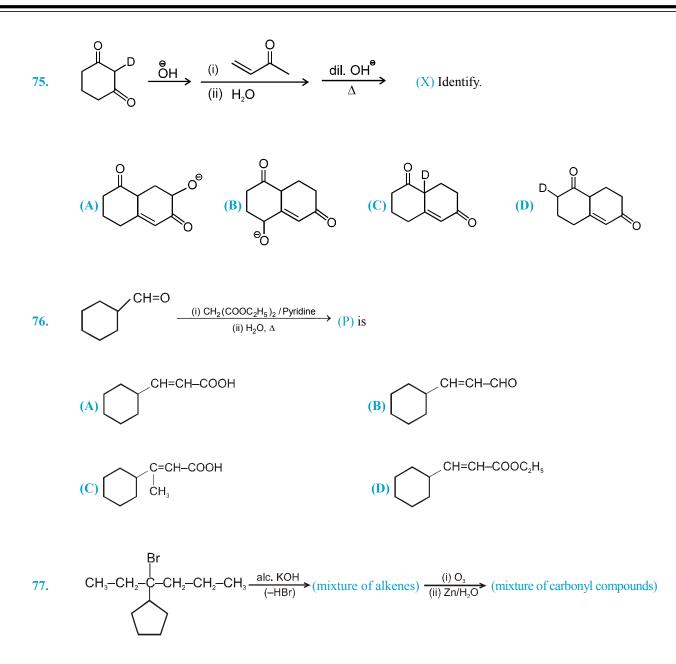
72. 
$$CI - \bigcirc -CHO + CH \swarrow CN \longrightarrow Product$$
(A) 
$$CI - \bigcirc -CH = C \swarrow CN$$
(B) 
$$CI - \bigcirc -CH - CH \swarrow CN$$
(C) 
$$OHC - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(E) 
$$CI - \bigcirc -CH \checkmark CN$$
(E) 
$$CI - \bigcirc -CH \land CN$$
(E) 
$$CI - \bigcirc -CH \land$$

A and B respectively are :

(A) 
$$\bigcirc$$
 + CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>

∠CN



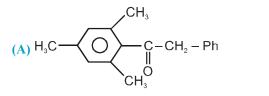


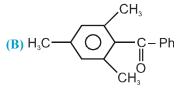
The incorrect statement is

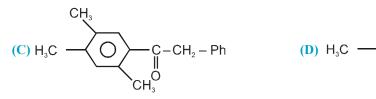
- (A) Total five alkenes are obtained
- (B) Total six different carbonyl compounds are obtained on ozonolysis
- (C) All carbonyl compounds can give aldol reaction when treated with dil KOH
- (D) Only two carbonly compounds give positive iodoform test

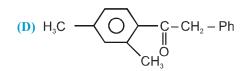
78. (i) 
$$H_3C \xrightarrow{CH_3} H_3C \xrightarrow{Mg/Ether} [A] \xrightarrow{(i)CO_2} [B] \xrightarrow{SOCl_2} [C] \xrightarrow{(i)} (C_6H_5CH_2)_2CuLi [D]$$

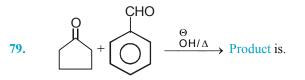
Identify (**D**) in the following sequence of reaction.

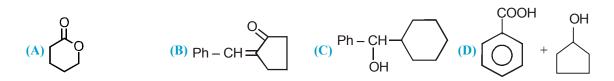




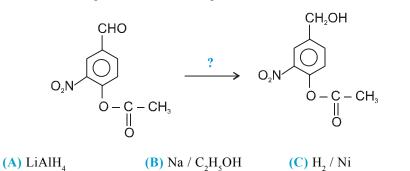


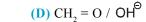






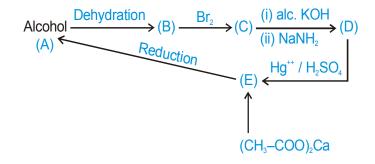
**80.** The suitable reagent for the following reaction is :





# SOLVED EXAMPLES

1. Find out unknown in following reactions.



Sol. Since E is obtained on dry distillation of calcium salt of acetic acid hence E will be CH<sub>3</sub>-C-CH<sub>3</sub>. Thus other unknowns are

$$\begin{array}{l} \mathsf{H} \\ \mathsf{A} = \mathsf{C}\mathsf{H}_{3} - \overset{\mathsf{C}}{\mathsf{C}} - \mathsf{C}\mathsf{H}_{3} \\ \mathsf{O}\mathsf{H} \end{array} \qquad \qquad \mathsf{B} = \mathsf{C}\mathsf{H}_{3} - \mathsf{C}\mathsf{H} = \mathsf{C}\mathsf{H}_{2} \\ \mathsf{O}\mathsf{H} \qquad \qquad \mathsf{D} = \mathsf{C}\mathsf{H}_{3} - \mathsf{C} = \mathsf{C} - \mathsf{H} \\ \mathsf{B}\mathsf{r} \quad \overset{\mathsf{H}}{\mathsf{B}\mathsf{r}} \qquad \qquad \mathsf{D} = \mathsf{C}\mathsf{H}_{3} - \mathsf{C} = \mathsf{C} - \mathsf{H} \\ \end{array}$$

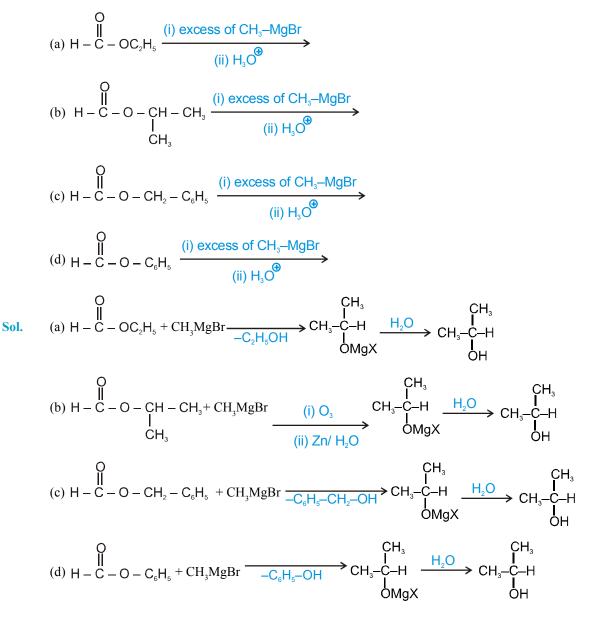
- 2. What will be structure of aromatic  $C_8H_8Cl_2$  (A), which on aqueous alkalihydrolysis gives product (B). (B) gives positive iodoform test.
- Sol. Since (B) is showing iodoform test hence it will be methylketone only as it is obtained on aqueous alkali hydrolysis of (A) which will be non-terminal gem dihalides as –

$$\begin{array}{ccc}
 & \mathsf{CI} & & \mathsf{KOH} \\
 & \mathsf{R} - \overset{\mathsf{C}}{\mathsf{C}} - \mathsf{CH}_3 & & \overset{\mathsf{KOH}}{\longrightarrow} & \mathsf{R} - \overset{\mathsf{C}}{\mathsf{C}} - \mathsf{CH}_3 & & \overset{\mathsf{KOH}}{\longrightarrow} & \mathsf{R} - \overset{\mathsf{O}}{\mathsf{C}} - \mathsf{ONa} + \mathsf{CHI}_3 \\
 & \mathsf{(A)} & & \mathsf{O} & & \mathsf{O} & \mathsf{O} & \mathsf{O} & \mathsf{O} \\
\end{array}$$

Now unknown 'R' can be known as :

$$R = C_8 H_8 C I_2 - C_2 H_3 C I_2 = C_6 H_5$$
Hence 'A' is  $C_6 H_5 - C_1 C H_3$ 

3. Write the products of the following reactions.



**4**. What will be hydration and hydroboration product for Cyclohexylethyne.

$$\bigcup_{Cyclohexylethyne} C C - H \xrightarrow{Hg^{2+} / H_2SO_4} \bigcup_{Cyclohexylmethylketone} C - CH_3$$

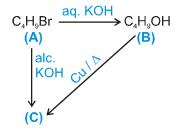
Sol. Cyclohexylethyne

(i) 
$$B_2H_6$$
 or  $Sia_2BH$   
(ii)  $H_2O_2$  (OH<sup>-</sup>)  
Cyclobexylethanal

- 5. Which hydrocarbon on ozonolysis gives acetone only ?
- **Sol.** Acetone only, means two moles of acetone.

$$CH_{3} - C = C - CH_{3} \xrightarrow{1. \text{ PCI}_{6}/\text{Ether}} A \xrightarrow{H_{2}O} B + C CH_{3} - C = O + O = C - CH_{3}$$
$$\underset{CH_{3}}{\overset{I}{\longrightarrow}} H_{2}O \xrightarrow{I} B + C CH_{3} - C = O + O = C - CH_{3}$$

6. Predict the structure of (A) in the following sequence :



- **Sol.** Since (B) is alcohol and (C) is alkene hence (B) is  $3^{\circ}$  alcohol only according to question (It is known that alkene can only be obtained from  $3^{\circ}$  alcohol when heated with copper). Thus structure of (B) is  $(CH_3)_3C$ -OH and its corresponding. alkyl bromide will be  $(CH_3)_3C$ -Br (tertiarybutylbromide)
- 7. What will be structure of  $C_4H_8O_2$  which on treating with excess  $CH_3$ -MgBr followed by acidification gives sole alcohol (A). (A) on treating with sodium hypoiodite solution gives positive iodoform test.
- Sol. Since (A) gives positively iodoform test hence it will be alkanol-2. 2° alcohol can be obtained only when alkylformate is treated with Grignard's reagent via aldehyde where alkyl part is alkyl part of Grignard's reagent. As Grignard's reagent is  $CH_3$ -MgBr hence 2° alcohol will be  $CH_3$ -CH-OH (propanol-2). Thus  $C_4H_8O_2$  is either  $CH_3$ -CH-OH (propanol-2).

$$\begin{array}{c} O & CH_3 \\ \parallel & \parallel \\ H-C-OCH_2-CH_2-CH_3 & \text{or} & H-C-O-CH-CH_3 \end{array}$$
Reactions:

Here we get two alcohols propanol-2 and propanol-1. Alkyl part of formic acid ester which gives propanol-2 will

be isopropyl only. Thus structure of  $C_4H_8O_2$  is  $H - C - OCH (CH_3)_2$ 

$$Ph \qquad C = N \qquad \xrightarrow{OH} \qquad \underbrace{1. \operatorname{PCl}_{6}/\operatorname{Ether}}_{2. \operatorname{H}_{2}O} A \xrightarrow{\operatorname{H}_{3}O^{+}} B + C$$

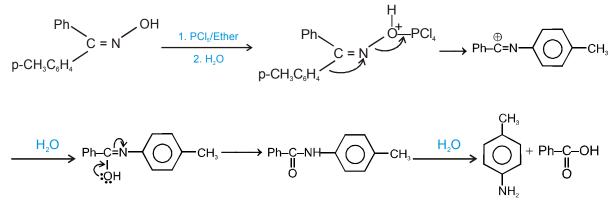
$$P-CH_{3}C_{6}H_{4} \qquad \underbrace{2. \operatorname{H}_{2}O} A \xrightarrow{\operatorname{H}_{3}O^{+}} B + C$$

$$A, B, C \text{ are}$$

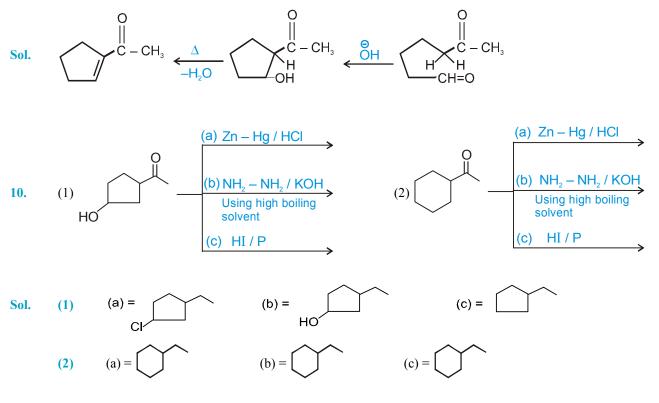
$$(A) \operatorname{PhCONH-p} - \operatorname{CH}_{3}C_{6}H_{4} \qquad (B) \operatorname{PhCOOH}$$

$$(C) \operatorname{pCH}_{3}C_{6}H_{4}\operatorname{NH}_{2} \qquad (D) \operatorname{PhCHO}$$

Sol. (A,B,C)



9. Which carbonyl compound on heated with dilute alkali gives 1- acetylcyclopentene.



- 11. (A) on treating with (B) in the presence of dry ether gives (C) which on acids hydrolysis gives (D). (D) on oxidation gives 2,5-dimethylhexan-3-one.
- Sol.

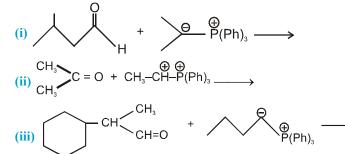
By knowing structure of given product (D) will be  $CH_3 - CH - CH_2 - CH - CH - CH_3$  hence (C) will be  $\begin{vmatrix} I \\ I \\ CH_3 \end{vmatrix}$  OH  $CH_3$ 

 $CH_3 - CH - CH_2 - CH - CH_3$  and finally A & B will have following two structures.  $| I - CH_3 = 0$   $CH_3 = 0$  $CH_3 = 0$ 

$$A = CH_3 - CH - CH_2 - CH = O \quad or \quad A = CH_3 - CH - CH = O$$

$$\begin{array}{ccc} B = \mathsf{CH}_3 - \mathsf{CH} - \mathsf{MgBr} & \text{or} & B = \mathsf{CH}_3 - \mathsf{CH} - \mathsf{CH}_2 - \mathsf{MgB} \\ & & & & \\ \mathsf{CH}_3 & & & \mathsf{CH}_3 \end{array}$$

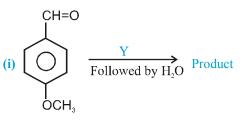
**12.** Predict the product for the followings :

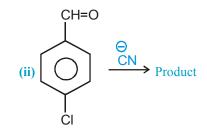


**Sol.** Witting reaction

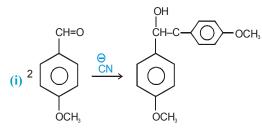


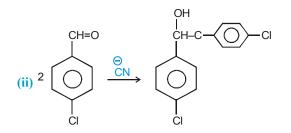
**13.** Predict Product –





Sol. Benzoin condensation reaction



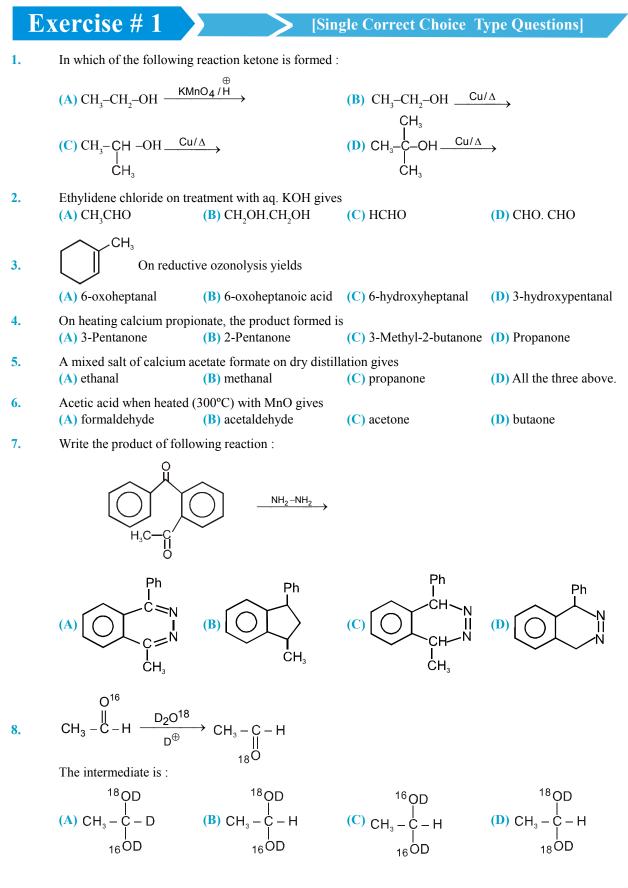


**14.** Predict product for the following

$$C_{6}H_{5} C = O \xrightarrow[C_{6}H_{5} - C - O - H]{H_{5} - C - O - H} Product$$

**Sol.** Baeyer-villiger oxidation

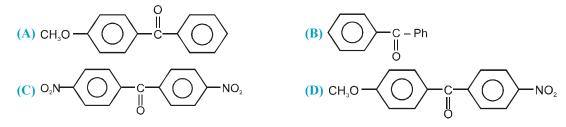
$$CH_{3}-C-OC_{6}H_{5} + C_{6}H_{5} - C - OH$$



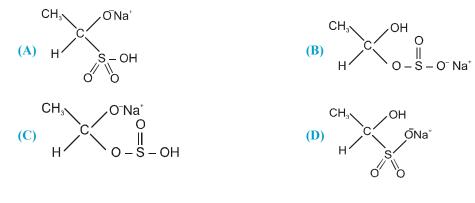
- 9. Aromatic carbonyl compounds having molecular formula C<sub>8</sub>H<sub>8</sub>O react with NH<sub>2</sub>OH how many oximes can be formed :
  - (A) 8 (B) 10 (C) 12 (D) 6
- 10.Arrange the following compounds in decreasing orders of rate of exchange of  $O^{18}$  with  $H_2O^{18}$ (X) CCl\_3CHO(Y) CH\_3CHO(Z) CH\_3COCH\_3(W) CF\_3CHO

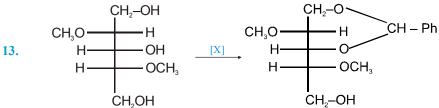
(A) W > Z > X > Y (B) W > X > Y > Z (C) W > Y > Z > X (D) W > Z > Y > X

11. Which of the following compound has the largest equilibrium constant for the addition of water ?



12. Acetaldehyde on reaction with sodium hydrogen sulphite produces



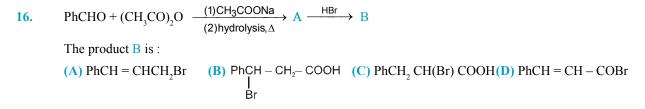


Compound (X) in the above reaction.

14.

The correct order of reactivity of PhMgBr with following compounds will be.							
$(C_6H_5)_2CO,$	CH	$CH_3 - CH = O$ ,		$(CH_3)_2 C = O$			
(1)	(2)		(3)				
(A) $1 > 2 > 3$	<b>(B)</b> $2 > 3 > 1$	(C) $3 > 2 > 1$		<b>(D)</b> $1 > 3 > 2$			

15.The cyanohydrin of a carbonyl compound on hydrolysis gives lactic acid. The carbonyl compound is(A) HCHO(B) CH<sub>3</sub>CHO(C) CH<sub>3</sub>COCH<sub>3</sub>(D) CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>



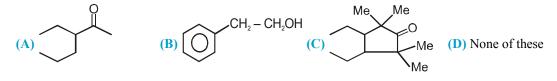
17. In the given reaction the product is :

$$CH_{3}-CH_{2}-C-OCH_{3} + CH_{3}-C-OCH_{3} \frac{(1) CH_{3}O^{\Theta} Na^{*}}{(2) CH_{3}OH} \xrightarrow{H_{2}O}{\Delta}$$

$$(A) CH_{3}-CH_{2}-C-CH_{2}-C-OCH_{3} \qquad (B) CH_{3}-C-CH-COOCH_{3}$$

$$(C) H-C-CH_{2}-CH-COOCH_{3} \qquad (D) CH_{3}-C-CH_{2}-CH_{3}$$

**18.** The compounds that undergo Aldol condensation is :

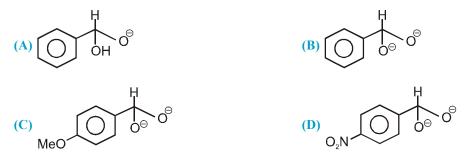


- 19.Which of the following will not undergo aldol condensation ?(A) CH<sub>3</sub>CHO(B) CH<sub>3</sub>CH<sub>2</sub>CHO(C) CD<sub>3</sub>CHO(D) PhCHO
- 20. (X) is the product of cross aldol condensation between benzaldehyde ( $C_6H_5CHO$ ) and acetone What is its structure ?

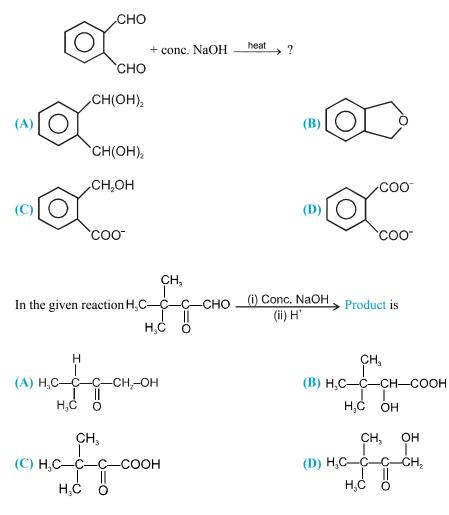
$$\begin{array}{c} O \\ \parallel \\ (A) C_6H_5-CH=CH-C-CH_3 \\ (C) C_6H_5-CO-CH_2-C=(CH_3)_2 \\ \end{array}$$

$$\begin{array}{c} O \\ (B) C_6H_5-CH=C-(CH_3)_2 \\ (D) \text{ None of these} \\ \end{array}$$

- In which of the following compounds the methylene hydrogens are the most acidic ?
   (A) CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>
   (B) CH<sub>3</sub>CH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub>
   (C) CH<sub>3</sub>CH<sub>2</sub>CH(COOC<sub>2</sub>H<sub>5</sub>)
- (D) CH<sub>3</sub>COCH<sub>2</sub>CN.
- 22. In the cannizzaro's reaction the intermediate that will be the best hydride donor ?

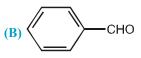


23. Product of following reaction is



25. Cannizzaro reaction does not take place with

(A) (CH<sub>3</sub>)<sub>3</sub>CCHO.

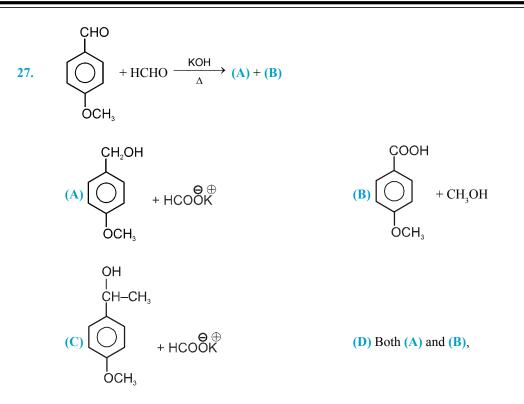


С)

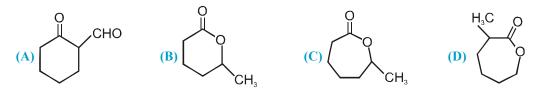
26. In the reaction,  $(CH_3)_3CCHO + HCHO \xrightarrow{\text{NaOH}} A + B$ .

the products (A) and (B) are respectively :

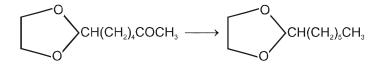
- (A)  $(CH_3)_3CCH_2OH$  and  $HCOO^-$  Na<sup>+</sup>.
- (B) (CH<sub>3</sub>)<sub>3</sub>CCOONa and CH<sub>3</sub>OH.
- (C) (CH<sub>3</sub>)<sub>3</sub>CCH<sub>2</sub>OH and CH<sub>3</sub>OH.
- **(D)**  $(CH_3)_3$ COONa and HCOO<sup>-</sup> Na<sup>+</sup>.



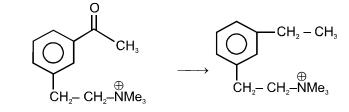
28. 2-Methyhlcyclohexanone is allowed to react with metachloroperbenzoic acid. The major product in the reaction is



**29.** In the following conversion

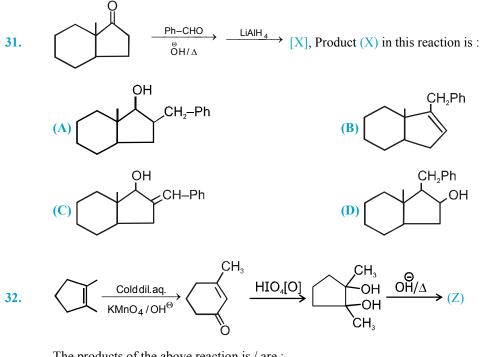


Which of the following regents is suitable ?(A) NH2NH2,KOH, DMSO(B) NaBH4(C) Zn-Hg, concentrated H2SO4(D) LiAlH4

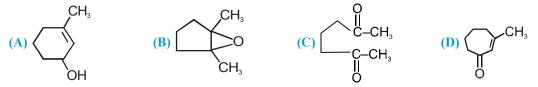


Above conversion can be achieved by (A) NH<sub>2</sub>-NH<sub>2</sub>/ NaOH (B) Zn-Hg/HCl (C) LiAlH<sub>4</sub>

**(D)** NaBH<sub>4</sub>.



The products of the above reaction is / are :

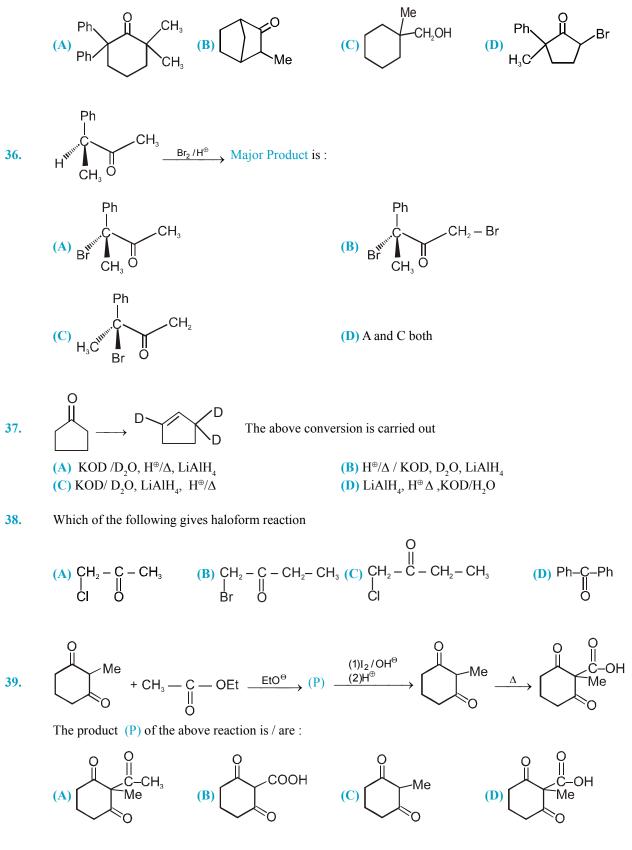


33. What will be the product of the following reaction

(A) Ph-CH-O-C-Me (B) Ph C -CH 
$$\stackrel{\text{RCO}_{3}H}{\underset{\text{Me}}{\overset{\text{RCO}_{3}H}{\overset{\text{RCO}_{$$

In which of the following reaction deuterium exchange is observed ? 34.

(A) 
$$CH_{3} - C - CH_{2} - CH_{3} \xrightarrow{H_{1}O/OH^{\Theta}} CD_{3} - C - CD_{2} - CH_{3}$$
  
(B)  $CH_{3} - CH - C - H \xrightarrow{D_{2}O/OH} CH_{3} - CD - C - H$   
Ph O  
(C)  $\xrightarrow{P_{1}} D_{2}O/OH \xrightarrow{P_{2}} D \xrightarrow{P_{1}} H$   
(D)  $H - C - H \xrightarrow{D_{2}O/OH} D - C - D$ 



In which of the following reaction deuterium exchange is not observed ?

40. 2-pentanone can be distinguished from 3- pentanone by the reagent ?
(A) 2, 4- Dinitrophenyl hydrazine
(B)Tollen's reagent
(C) I<sub>2</sub> and dilute NaOH
(D) NaHSO<sub>3</sub>

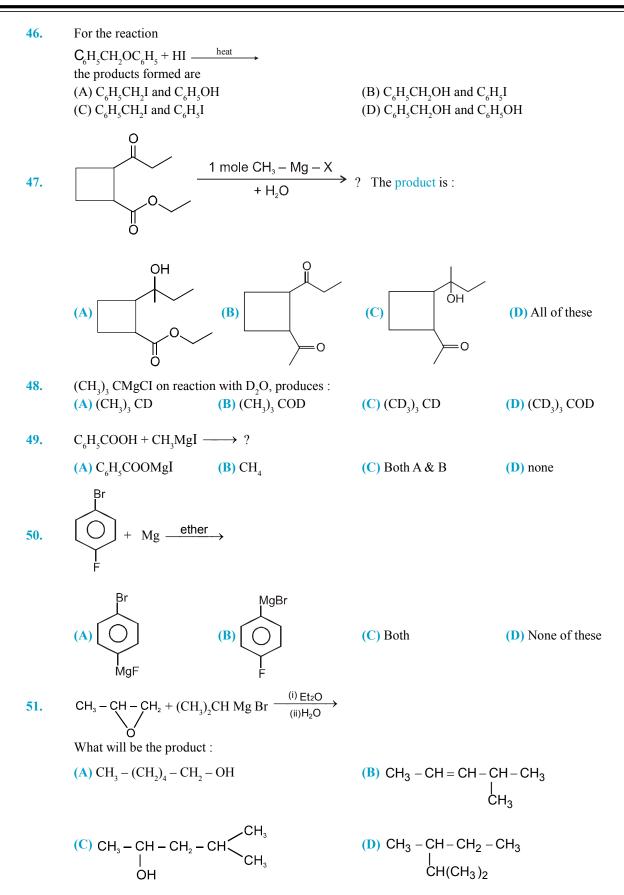
41. 
$$\begin{array}{c} & \bigoplus_{i=1}^{Q} + (C_{6}H_{5})_{3}P = CHCH_{3} \longrightarrow \\ & \bigoplus_{i=1}^{Q} - CH_{3} & \bigoplus_{i=$$

$$C_{6}H_{5}CHO + CH_{3}NO_{2} \xrightarrow{\text{NaOH}} (X)$$
(A)  $C_{6}H_{5} - CH - CH - NO_{2}$ 
(B)  $C_{6}H_{5}CHCH_{2}NO_{2}$ 
(C)  $C_{6}H_{5}CH = CH - NO_{2}$ 
(D)  $C_{6}H_{5}CH \xrightarrow{CH_{2}NO_{2}} CH_{2}NO_{2}$ .

**43.** What is the final product of this sequence of reactions ?

44. Ph-CH<sub>2</sub>-COOEt + 
$$\begin{array}{c} O\\ EtO\end{array}$$
 (B) Ph-CH<sub>2</sub>-COOEt (C) Ph-CH (D) None of these (D) None of these

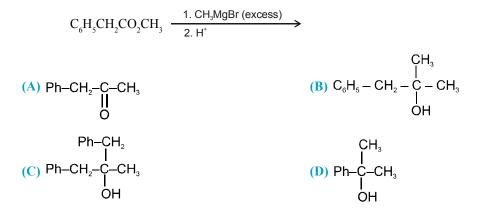
45. Which of the following will gives iodoform with NaOI ?



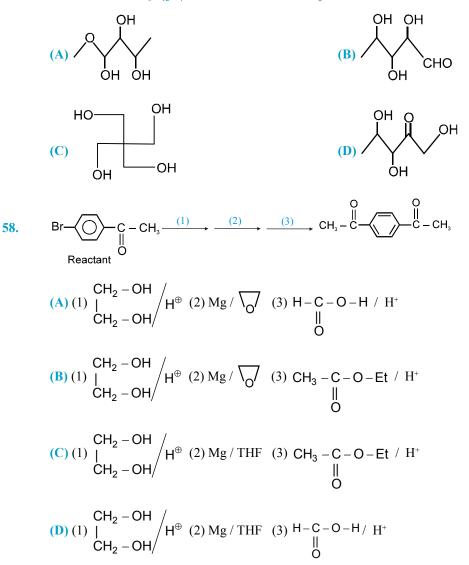
54. How many functional group produced  $CH_4$  gas by the reaction of compound (I) with  $CH_3MgBr$ .

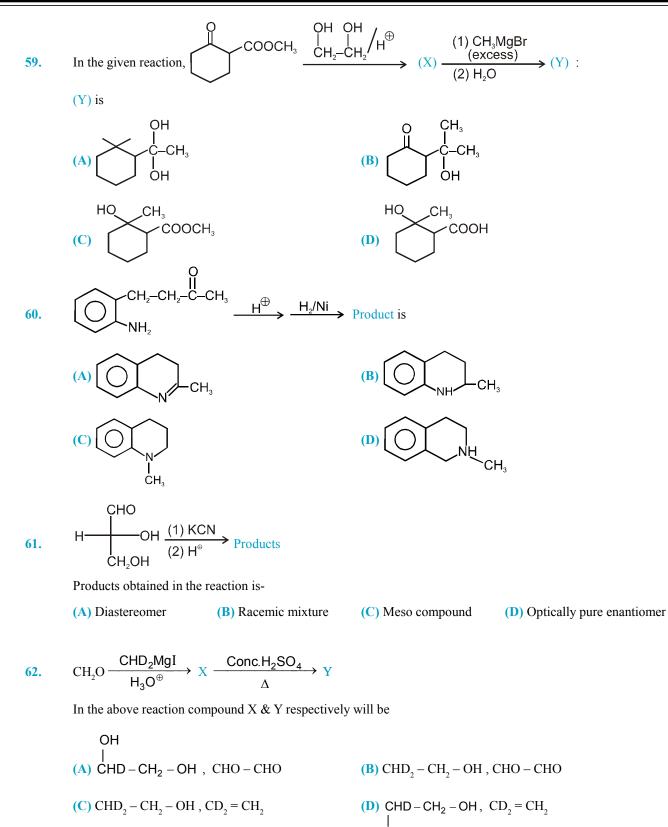
OH HO--CH₃ οŢ\_\_\_ SO<sup>®</sup>H HOOC-۰N **(I**) **(A)** 3 **(B)** 4 **(C)** 5 **(D)** 6  $\begin{array}{c} O \\ \parallel \\ Ph-MgBr + (CH_3)_2CH - C - CI \longrightarrow [X] \xrightarrow{Zn-Hg/HCI} [Y] \end{array}$ 1 mole Identify structure of [Y].  $\begin{array}{c} \textbf{(B)} \ \mathsf{Ph}-\mathsf{CH}-\mathsf{CH}_2-\mathsf{CH}_3\\ & \downarrow\\ \mathsf{CH}_3 \end{array}$ (A)  $Ph - CH_2 - CH - CH_3$ | CH3  $(C) Ph-C-CH_3 \\ | \\ CH_3 \\ CH_3$ (**D**)  $\bigcirc$   $-CH_2 - CH_2 - CH_3$  $CH_3$ 

56. Predict the major product in the following reaction:



57. A compound X  $(C_5H_{12}O_4)$  upon treatment with CH<sub>3</sub>MgX gives 4 mole of methane. Identify the structure of (X).



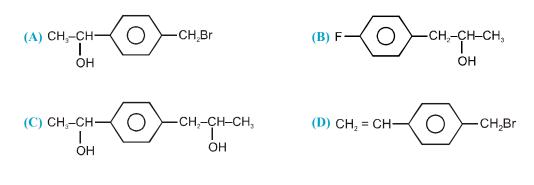


ÒН

234

63. For the given reaction 
$$F \longrightarrow CH_2Br + Mg \xrightarrow{dry \text{ ether}} A \xrightarrow{(i)CH_3CHO} B$$

product **B** is :

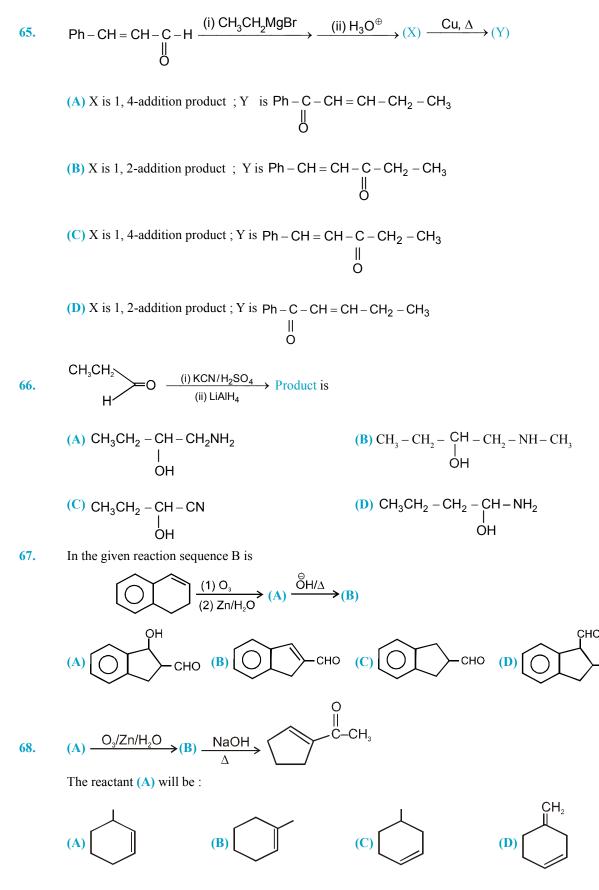


64. Consider the following sequence of reactions-.

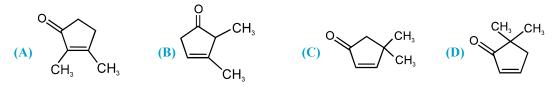
$$PhC \equiv CH \xrightarrow{HgSO_4} A \xrightarrow{NH_2OH} B + C.$$

The products (A), (B) and (C) are respectively,

(A) PhCHO,  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$  and  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$ (B) PhCH<sub>2</sub>CHO,  $\stackrel{PhCH_2}{\longrightarrow} C = N \quad OH$  and  $\stackrel{PhCH_2}{\longrightarrow} C = N \quad OH$ (C)  $\stackrel{Ph}{\longrightarrow} C = O$ ,  $\stackrel{Ph}{\longrightarrow} C \stackrel{OH}{\longrightarrow} OH$  and  $\stackrel{Ph}{\longrightarrow} C = N - OH$ (D)  $\stackrel{Ph}{\longrightarrow} C = O$ ,  $\stackrel{Ph}{\longrightarrow} C = N \quad OH$  and  $\stackrel{Ph}{\longrightarrow} C = N - OH$ 

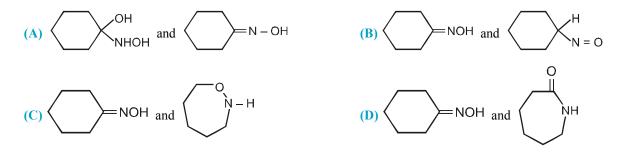


69. 
$$CH_3 \xrightarrow{\text{KOH, H}_2O} Product (C_7H_{10}O) :$$
  
 $CH_3 \xrightarrow{\text{CH}_3} CH_3 \xrightarrow{\text{KOH, H}_2O} Product (C_7H_{10}O) :$ 



70. Consider the following sequence of reactions :

The products (A) and (B) are, respectively :



71. Compound (X)  $C_9H_{10}O$  gives yellow coloured ppt with 2,4 DNP but does not give red coloured ppt with Fehling's solution. (X) on treatment with  $NH_2OH/H^+$  gives compound (Y)  $C_9H_{11}NO$ . (Y) when treated with  $PCl_5$  gives isomeric compound (Z). (Z) on hydrolysis gives propanoic acid and aniline. What will be the correct structure of (X), (Y) and (Z) ?

(A) 
$$C_{6}H_{5} - C - C_{2}H_{5};$$
  $C_{6}H_{5} - C - C_{2}H_{5};$   $C_{6}H_{5} - C - C_{2}H_{5};$   $C_{6}H_{5} - C - NH - C_{2}H_{5}$   
 $|| O OH - N O$ 

(C) 
$$C_6H_5 - CH_2 - C - CH_3$$
;  $C_6H_5 - CH_2 - C - CH_3$ ;  $CH_3 - C - CH_2 - NH - C_6H_5$   
 $\| \\ O \\ N - OH \\ O \\ O$ 

(D) 
$$C_6H_5 - C - C_2H_5$$
;  $C_6H_5 - C - C_2H_5$ ;  $C_6H_5 - C - NH - C_2H_5$   
 $|| \\ O \\ N - OH \\ || \\ O$ 

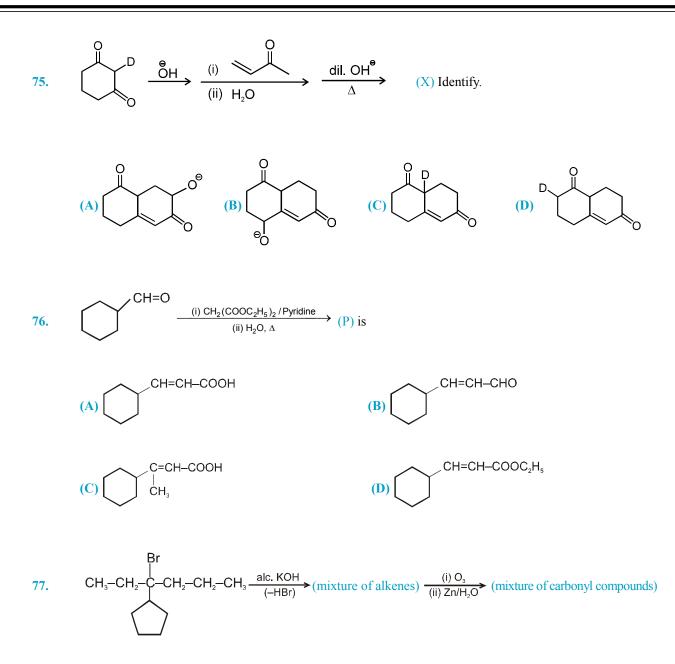
72. 
$$CI - \bigcirc -CHO + CH \swarrow CN \longrightarrow Product$$
(A) 
$$CI - \bigcirc -CH = C \swarrow CN$$
(B) 
$$CI - \bigcirc -CH - CH \swarrow CN$$
(C) 
$$OHC - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(D) 
$$CI - \bigcirc -CH \swarrow CN$$
(E) 
$$CI - \bigcirc -CH \checkmark CN$$
(E) 
$$CI - \bigcirc -CH \land CN$$
(E) 
$$CI - \bigcirc -CH \land$$

A and B respectively are :

(A) 
$$\bigcirc$$
 + CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>

∠CN



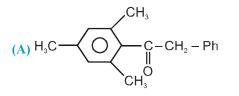


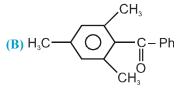
The incorrect statement is

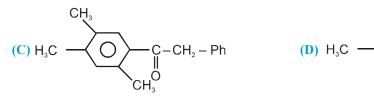
- (A) Total five alkenes are obtained
- (B) Total six different carbonyl compounds are obtained on ozonolysis
- (C) All carbonyl compounds can give aldol reaction when treated with dil KOH
- (D) Only two carbonly compounds give positive iodoform test

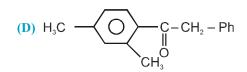
78. (i) 
$$H_3C \xrightarrow{CH_3} \xrightarrow{Mg/Ether} [A] \xrightarrow{(i)CO_2} [B] \xrightarrow{SOCl_2} [C] \xrightarrow{(i) (C_6H_5CH_2)_2CuLi} [D]$$

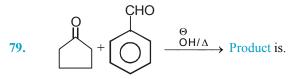
Identify (**D**) in the following sequence of reaction.

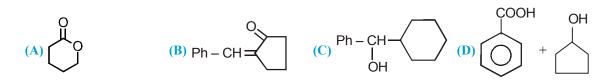












**80.** The suitable reagent for the following reaction is :

240

