

# **SOLVED EXAMPLES**

**Ex.1** How may 1°, 2°, 3° and 4° carbon atoms are present in following molecule.

$$\begin{array}{c} CH_{3} \\ CH_{3} - CH - CH_{2} - CH_{2} - CH_{-}CH_{3} \\ CH_{3} \\ CH_{3} \\ CH_{3} \\ CH_{3} \\ CH_{2} - CH_{3} \end{array}$$

- **Sol.**  $1^{\circ}$  Carbon atoms = 6,  $2^{\circ}$  Carbon atoms = 2,  $3^{\circ}$  Carbon atoms = 2,  $4^{\circ}$  Carbon atom = 1
- **Note**: Primary, secondary, tertiary & quaternary carbon atoms in a molecule are denoted by the letters p, s, t and q respectively.
- **Ex.2** How many 1°, 2°, 3° and 4° carbon atoms are present in following molecule.

$$CH_{3} - CH_{2} - CH_{2} - CH_{3}$$

$$CH_{3} - CH_{3} - CH_{3}$$

$$CH_{3} - CH_{3}$$

$$CH_{3} - CH_{3}$$

Sol. 
$$\overset{l^{\circ}}{\overset{l^{\circ}}{\operatorname{CH}_{3}}} - \overset{2^{\circ}}{\overset{c}{\operatorname{CH}_{2}}} - \overset{3^{\circ}}{\overset{c}{\operatorname{CH}_{3}}} - \overset{l^{\circ}}{\overset{l^{\circ}}{\operatorname{CH}_{3}}} + \overset{l^{\circ}}{\overset{l^{\circ}}{\operatorname{CH}_{3}}} +$$

 $1^{\circ}$  Carbon atoms = 5,  $2^{\circ}$  Carbon atom = 1,

 $3^{\circ}$  Carbon atom = 1,  $4^{\circ}$  Carbon

 $4^{\circ}$  Carbon atom = 1

**Ex.3** Write the IUPAC name of following compounds. (i)  $H_3C-CH_2-CH-COOH$ 

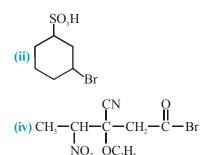
CH,

(ii) 3-Bromocyclohexane-1-sulphonic acid

(iv) 3-Cyano-3-ethoxy-4-nitropentanoyl bromide

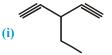
**Sol.** (i) 2-Ethoxybutanoic acid

(iii)

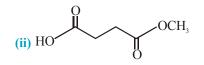


(iii) 1,1,2-Trimethylcyclopentane

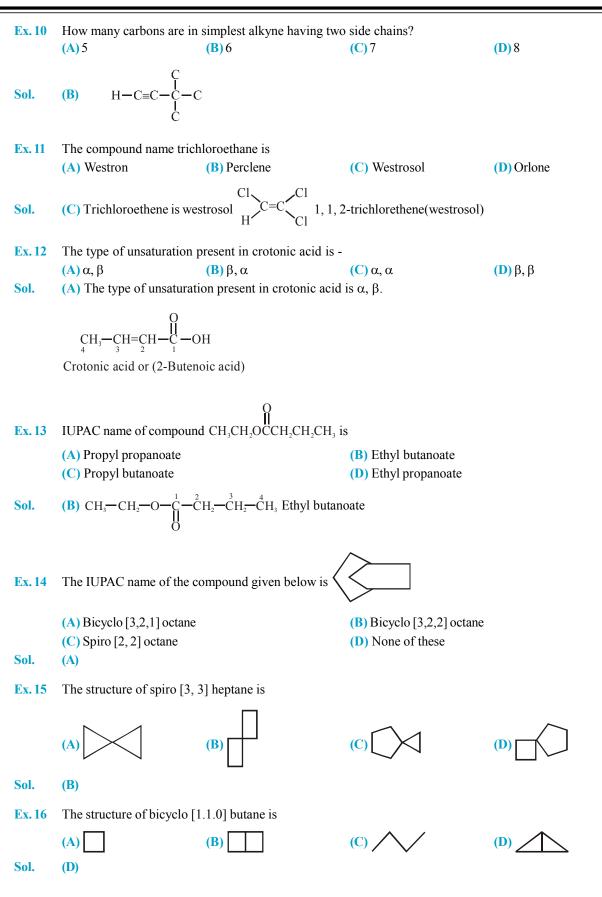
**Ex.4** Draw the structure of following IUPAC name.



(ii) 3-Methoxycarbonylpropanoic acid



**Sol.** (i) 3-Ethypenta-1,4-diyne



- 14. The IUPAC name of the following is [CH<sub>3</sub>CH(CH<sub>3</sub>)]<sub>2</sub>C(CH<sub>2</sub>CH<sub>3</sub>)C(CH<sub>3</sub>)C(CH<sub>2</sub>CH<sub>3</sub>)<sub>2</sub>
  (A) 3,5-Diethyl-4,6-dimethyl-5-[1-methylethyl]hept-3-ene
  (B) 3, 5-Diethyl-5-isopropyl-4, 6-dimethylhept-2-ene
  (C) 3,5-Diethyl-5-propyl-4, 6-dimethylhept-3-ene
  (D) None of these
- 15. Which of the following is a heterocyclic compound

$$\begin{array}{c} \text{HC=CH} \\ \text{(A)} \\ \text{HC=CH} \\ \text{HC=COOH} \\ \end{array} \begin{array}{c} \text{HC=COOH} \\ \text{HC=COOH} \\ \text{HC=COOH} \\ \end{array}$$

$$(C) \stackrel{HC=CH}{\underset{HC=CH}{\overset{I}{\xrightarrow{}}}} CH_2 \qquad (D) \stackrel{HC=CH}{\underset{HC=CH}{\overset{I}{\xrightarrow{}}}} C=O$$

**16.** Ethyl methyl vinyl amine has the structure –

(A) 
$$CH_3CH_2$$
— N—  $CH_2CH=CH_2$   
 $I$   
 $CH_3$   
(C)  $CH_2=CH$ — N—  $CH=CH_2$   
 $I$   
 $CH_2$ 

- 17.  $CH_3-CH=CH-C=CH$ , IUPAC name is : (A) Pent-2-ene-4-yne (B) Pent-4-yne-2-ene
- 18. The IUPAC name of  $CH_3 C \equiv C C(CH_3)_3$  is : (A) Methyl tertirarybutyl acetylene (C) 4, 4-Dimethyl-2-pentyne
- **19.** Give the IUPAC name of

$$CH_3$$
  
 $H_3C - C - CH_3$   
 $CH_3 - CH_2 - CH_2 - CH - CH - CH_2 - CH_2 - CH_3$   
 $H_3C - CH_4$   
 $H_3C - CH_5$   
 $H_3C - CH_5$   

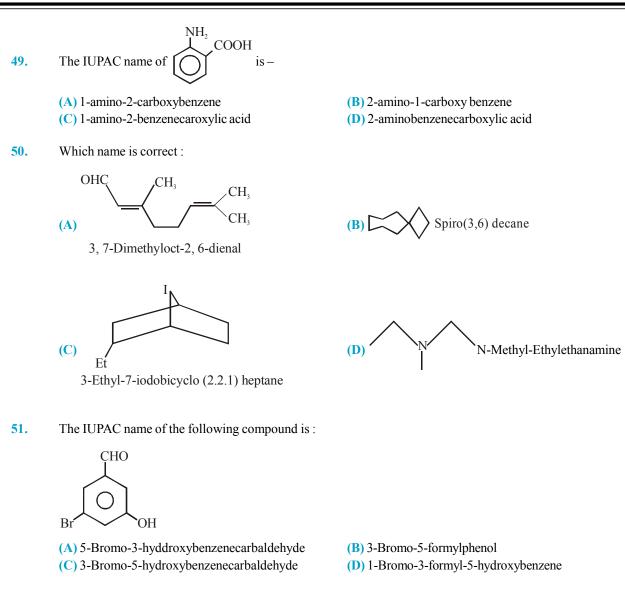
(A) 4-isopropyl-5-ter. butyl octane(C) 2-methyl-3-propyl-4-ter. butyl heptane

- (B)  $CH_3CH_2 N CH = CH_2$  I  $CH_3$ (D)  $CH_3 - N - CH = CH_2$  I $CH_3$
- (C) Pent-1-yne-3-ene (D) Pent-3-ene-1-yne
- (B) t-Butyl propyne(D) 1, 3, 3, 3 Tetramethyl ethyne

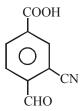
(B) 4-ter. butyl-5-isopropyl octane(D) 2, 2-dimethyl-3-propyl-4-isopropyl heptane

- 20. As per IUPAC rules, which one of the following groups, will be regarded as the principal functional group ? (A)  $-C \equiv C -$  (B) -OH (C) -C - (D) -C -HII O
- 21.The number of C-atoms in second member of an ester is/are :(A) 2(B) 3(C) 4(D) 1
- 22.The number of primary, secondary and tertiary carbon atom in toluene is given by the set :(A) 1, 6, 0(B) 1, 5, 1(C) 2, 5, 0(D) 1, 6, 1
- 23. C<sub>3</sub>H<sub>6</sub>Br<sub>2</sub> can shows :
  (A) Two gem dibromide
  (C) Two tert. dibromo alkane
  (B) Three vic dibromide
  (D) Two sec. dibromo alkane

33.	The IUPAC name of $\stackrel{\text{Et Me}}{\longmapsto}$ is :							
55.								
	(A) 2, 3-Dimethyl hexane	(B) 2-Ethyl-4-methyl pentane						
	(C) 3-Ethyl-2-methyl pentane	( <b>D</b> ) 2, 4-Dimethyl hexane						
34.	The IUPAC name of the compound is $CH_3 - CH - CH - NH_2$							
	CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>							
	(A) 1-Amino-1-phenyl-2-methyl propane	(B) 2-Methyl-1-phenyl propane-1-amine						
	(C) 2-Methyl-1-amino-1-phenyl propane	(D) 2-Chloro-2-Methylpropane						
35.	The IUPAC name of the compound Br (Cl) CH.CF <sub>3</sub> is :							
	(A) haloethane	(B) 1, 1, 1-triflouro-2-bromo-2-chloroethane						
	(C) 2-bromo-2-chloro-1, 1, 1-triflouroethane	<b>(D)</b> 1-bromo-1-chloro-2, 2, 2-triflouro ethane						
36.	IUPAC name of compound is CH <sub>3</sub> —CH—CH <sub>2</sub> —Cl	$H(OH) - CH_3$						
	IUPAC name of compound is $CH_3$ — $CH$ — $CH_2$ — $CI$ I $CH_2CH_3$							
	(A) 4-methyl-3-hexanol (B) heptanol	(C) 4-methyl-2-hexanol (D) none of these						
37.	The IUPAC name of tert-butyl chloride is :							
	(A) 4-Chlorobutane	(B) 2-Ethyl-2-methyl pentane						
	(C) 3-Ethyl-2-methyl pentane	(D) 2-Chloro-2-Methyl propane						
•	The IUPAC name of $$ is –							
38.	The IUPAC name of $\sqrt{15}$ (A) 4-ethyl-3-methyl hexane	(B) 3-ethyl-4-methyl hexane						
	(C) 3-methyl-4-ethyl hexane	(D) None of these						
39.								
37.	The correct nomenclature (IUPAC) for the following alcohol is :							
	CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub>							
	$CH_3CH_2$ OH (A) 2-Ethyl-2-butanon	<ul><li>(B) 1-Ethyl-1-methyl-pentanol-1</li><li>(D) diethyl ethanol</li></ul>						
	(C) 3-Ethyl pentan-3-ol							
40.	The IUPAC name of is :							
40.	(A) 1, 1-diethyl-2, 2-dimethyl pentane	(B) 4, 4-dimethyl-5, 5-diethylpentane						
	(C) 5, 5-diethyl-4, 4-dimethylpentane	(D) 3-ethyl-4, 4-dimethylheptane						
41.	Underline carbon is sp <sup>3</sup> hybridised in :							
	(A) $CH_3 - CH = CH_2$	<b>(B)</b> $CH_3CH_2 - NH_2$						
	(C) $CH_3CONH_2$	(D) CH <sub>3</sub> CHCN						
42.	The IUPAC name of							
	(A) 2-ethyl-3-methyl-1-penten-4-yne	(B) 2-ethyl-3methyl-4-pentyn-1-ene						
	(C) 4-ethyl-3-methyl-1-pentyn-4-ene	( <b>D</b> ) 4-ethyl-3-ethyl-4penten-1-yne						
70								



52. The IUPAC name of the compound is :



(A) 2-Cyano-1-fromylbenzene-4-carboxylic acid (C) 4-Careboxy-2-cyanobenzene-1-carbaldehyde

53. IUPAC name of Cl 
$$C$$
  $C$   $C$ 

(A) 4-Chlorophenyl benzoate,(C) Benzyl-4-chlorobenzenecarboxylate

(B) 3-Cyano-4-formylbenzene-1-carboxylic acid (D) 2-Formyl-5-carboxybenzene-1-carbonitrile

(B) Phenyl-4-Chlorobenzenecarboxylate.(D) 4-Chloro diphenylcarboxylate.

#### Exercise # 2 Part # I [Multiple Correct Choice Type Questions]

- Which of the following statement is /are wrong? 1.
  - (A)  $C_{n}H_{2n}$  is the general formula of alkanes
  - (B) In homologous series, all members have the same physical properties
  - (C) IUPAC means International Union of Physics and Chemistry
  - (D) Butane contains two 1° C atoms and 2°C atom
- 2. Which of the following statement is/are correct?
  - (A) Homologous series can be represented by a general formula
  - (B) The chemical properties of an organic compound depend on the functional group
  - (C) Group obtained by the removal of one H atom from the alkane are called alkyl groups
  - (D) Alkynes consist of one double-bond in their molecules
- 3. Which of the following statement is/are correct?
  - (A) Methane was named as fire damp as it formes explosive mixture with air
  - (B) Primary suffixes are added to the root word to show saturation or unsaturation in a C atom
  - (C) The IUPAC name of the valeric acid is pentanoic acid
  - (D) The common name of hexanoic acid is caproic acid
- Which of the following statement is /are correct? 4.
  - (A) The IUPAC name of amyl alcohol is pentanol
  - (B) The IUPAC name of isoamyl alcohol is 3-methyl butanol
  - (C) Wood spirit is methanol
  - (D) Methyl alcohol is also called carbinol
- Which of the following statement is/are correct? 5.
  - (A) The trivial names of organic compounds are called common names
  - (B) The systematic names of organic compound are obtained from the IUPAC system
  - (C) The systematic name of alkanes are based on the number of C atom in the longest continues chain of C atoms

(D) The maximum number of functional groups must be included in the C atom chain selected even if it does not satisfy the longest chain rule

Which of the following statement is/are wrong? 6.

(A) Acetic acid is the systematic of vinegar

(B) Me-C-OH is an unsaturated compound

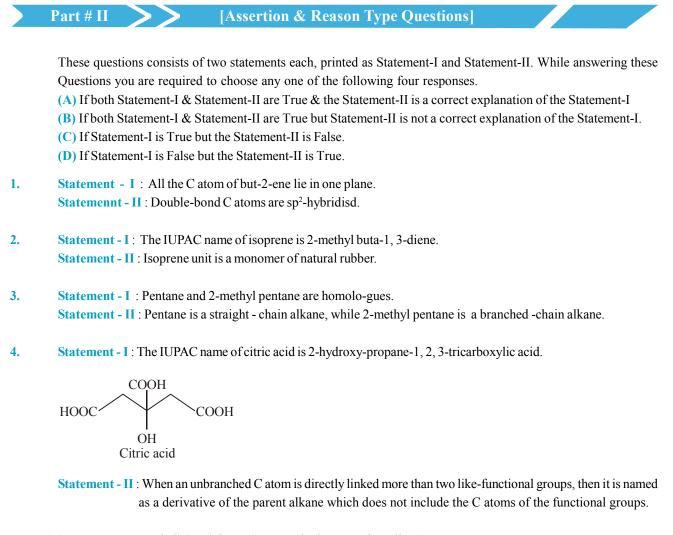
- (C) Prefixes like n-, iso, sec-, tert, neo- etc. are used in IUPAC system.
- (D) The systematic names of acids are formed by dropping –e of the name of parent alkane and adding –oic acid.
- Which of the following statement is /are correct? 7.

(A) R - C - O - C - R is an unsaturated compound

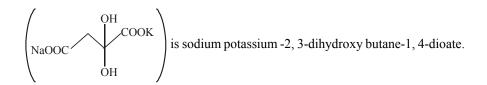
(B) Neohydrocarbons contain a 3° C atom

(C) The IUPAC name of isopropyl alcohol is propan-2-ol

(D) The IUPAC name of (CH<sub>2</sub>CN) is ethanenitrile



Statement - I : Rochelle's salt is used as complexing agent in Tollen's reagent.
 Statement - II : Sodium potassium salt of tartaric acid is known as Rochelle's salt. The IUPAC name of Rochelle's salt



5. Match the column Column - I Compound (A) **(p)** HO Vitamin D<sub>3</sub>  $NH_2$ Ph **(B)** Me **(q)** Amphetamine Me H **(C) (r)** A cockroach repellent found in cucumbers **(s)** Match the column 6. Column - I Compound Me **(A)** Me **(p)** Me Мe Me, Me **(B)** Me Me **(q)** Мe Me Me Me **(C)** Me **(r)** Me Me OH **(D)** Me Me **(s)** Me Me Match the column 7. Column - I Compound Ph Me **(A)** H,N **(p)** Aspartame Ph Me **(B) (q)** Demerol **(C) (r)** Asynthetic cockroach repellent **(s)** 

#### Column - II Containing all the functional groups

**p)** 1° amine

2° alcohol

) Triene

Aldehyde and ene

Column - II Nature of H atoms

(p)  $15(1^{\circ} H), 4(2^{\circ} H), 1(3^{\circ} H)$ 

**q)**  $17(1^{\circ}\text{H}), 2(2^{\circ}\text{H}), 2(3^{\circ}\text{H})$ 

(r) 12(1°H), 2(2°H), 2(3°H)

(s) 15(1°H), 2(2°H), 1(3°H)

Column - II Containing all the functional groups

ene and diester

Carboxylic acid, 1° amine, amide

) Ester

s) 3° amine

	CH <sub>2</sub> COOH	
3.	CH <sub>2</sub> COOH HOOC-CH <sub>2</sub> -CH-CH <sub>2</sub> -COOH	3-(carboxymethyl)-1, 5-dioic acid
	True (A)	False (B)
4.	CH <sub>2</sub> CHO I OHC-CH <sub>2</sub> -CH-CH <sub>2</sub> -CHO	3-(formylmethyl)pentane-1, 5-dial
	True (A)	False (B)
5.	$H_{2}NOC-CH_{2}-CH-CH_{2}-CONH_{2}$ True (A)	Propane-1, 2, 3-tricarboxamide False <b>(B)</b>

## **Comprehension #2**

In addition to the standard ring systems (such as cyclohexane), cyclic compounds can also be bicyclic, tricyclic, etc. or they can be spirocyclic, bicyclic or bridge head carbons. The point of attachment of two rings are called bridge head atoms.

The formal names of bicyclic and related ring systems are based on

(A) Total number of atoms in the molecule.

(B) The number of atoms in each bridge connecting the bridge head atoms. These numbers are written in square bracket in decreasing order.

Spirocyclic compounds have two fused rings, but only bridge head atom. Spirocyclic compounds are named like bicyclic compounds, but have the prefix spirocyclo. Answer the following question :

What is the IUPAC name of the above compound ? (A) cyclo [1.2.2] heptane (C) Bicyclo [2.2.1] heptane

(B) Bicyclo [1.2.2] heptane (D) cyclo [2.2.1] heptane

2.

The number of atoms in each bridge are :

**(B)**[3.1.0]

(A) [3.2.1]

**(C)**[1.3.0]

**(D)**[2.1.0]

3. Select the correct statement about the following compounds :

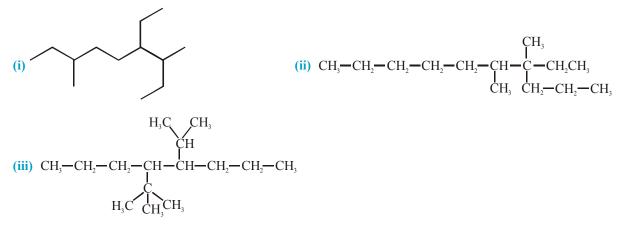
(A) It is a tricyclic compound (C) It is spiro compound

(B) It is bicyclo compound (D) Its IUPAC name is bicyclo [2.2.2] hexane

Which of the following is the correct structure of bicyclo [1.1.0] butane? 4.



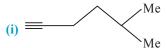
7. Write the correct IUPAC name of the following compounds.



8. Write IUPAC name of the following : -(i)  $(CH_3)_3C - CH = CH_2$ (iii)  $CH_3 - CH_2 - CH_2 - CH_2 - CH_2 - CH_2$ (iii)  $CH_3 - CH_2 - CH_2 - CH_2 - C = CH_2$ 

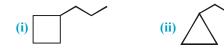
(ii) 
$$CH_2 = CH - CH = CH_2$$

9. Write IUPAC name of the following





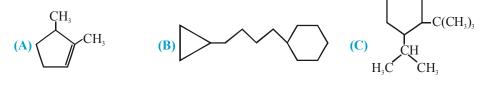
**10.** Write correct IUPAC name of the following



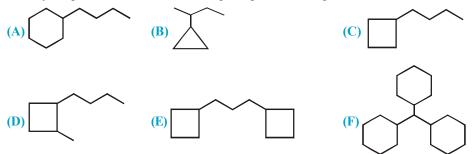




**11.** Write the IUPAC name of the following compounds.



12. Identify the parent chain in the following compounds as ring or side chain.



**19.** Write IUPAC name

(A) 
$$CH_3CH_2$$
-CH-C-OCH<sub>3</sub> (B)  $CH_3$ -CH-CH<sub>2</sub>-C-O-C<sub>2</sub>H<sub>5</sub> (C)  $CH_3$ -C-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_2$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_3$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_3$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_3$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_3$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>  
 $CH_3$  (C)  $CH_3$ -C-CH-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-C

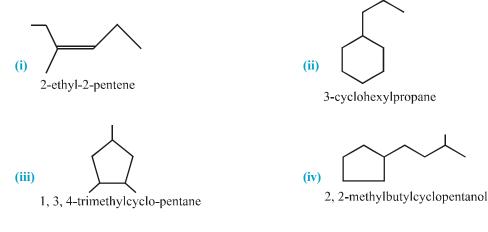
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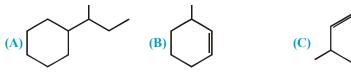
Write the correct IUPAC name of the following compounds.

(A) 
$$CH_3 - CH - CH_2 - COOH$$
  
 $CH_3 - CH_2 - CH - CH_2 - CH - CH_2 - CHO$   
(B)  $CH_3 - CH_2 - CH - CH_2 - CHO$   
 $CH_3 - CH_2 - CH - CH_2 - CHO$   
(C)  $CH_3 - CH_2 - CH_2 - CH_2 - CH$   
(D)  $CH_3 - CH - CH_2 - CH - CH_2 - CHO$ 

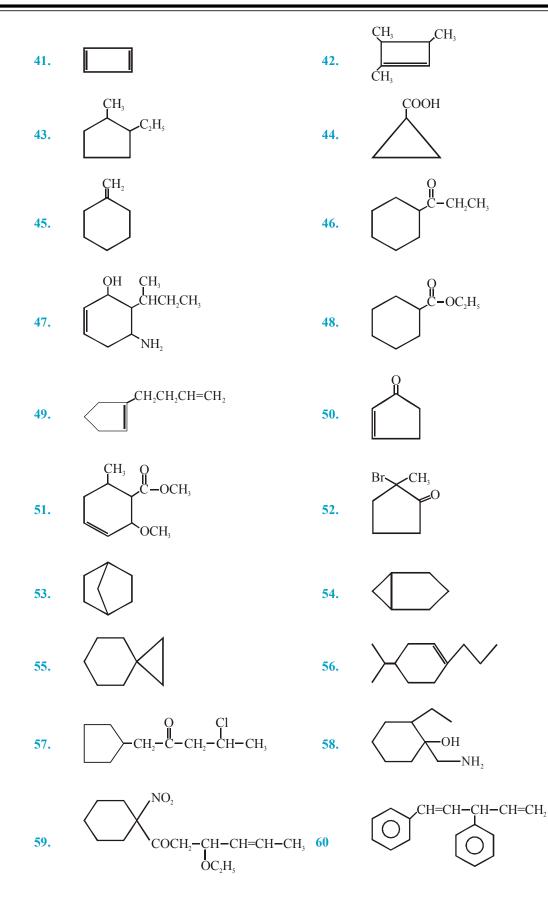
- 21. A certain substances contains only carbon and hydrogen and has a molecular weight of 70. Photochemical chlorination gave only one monochloride. Write the structure and IUPAC name of the hydrocarbon and its monochloride.
- 22. A hydrocarbon of molecular weight 72 g mol<sup>-1</sup> has a 2-methyl group. What is the IUPAC name? Also drawn its bond-line structure ?
- Write the structure and give IUPAC systematic name of an alkane or cycloalkane with the formula :
   (A) C<sub>8</sub>H<sub>18</sub> that has only primary hydrogen atoms
   (B) C<sub>6</sub>H<sub>12</sub> that has only secondary hydrogen atoms.
- 24. What is wrong with the names given for these compounds provide the correct name for each :



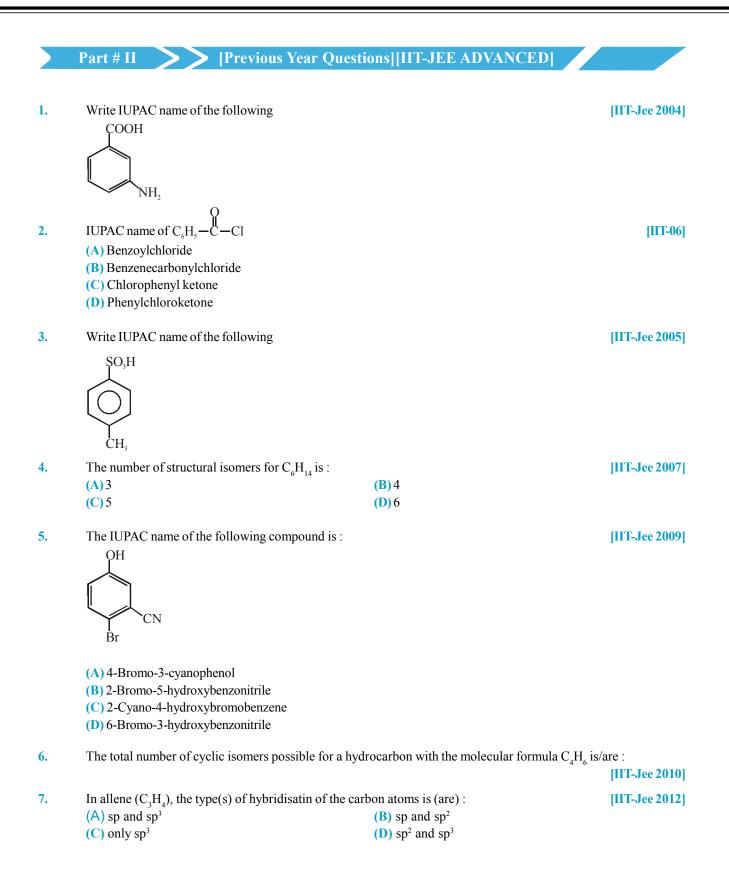
25. Write the IUPAC name for each the following structures :

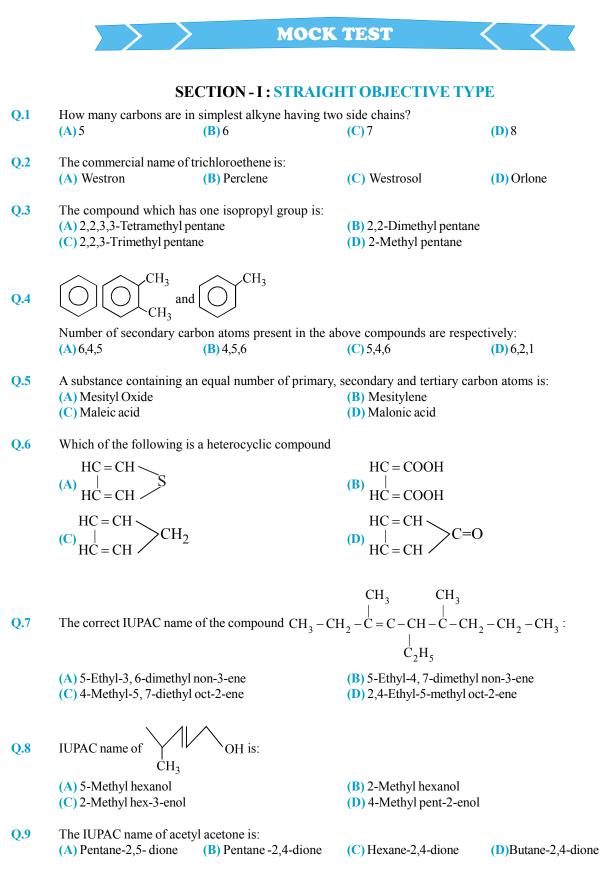


26. Write down the correct priority for citation as principal groups :



-NH<sub>2</sub>





PO-C=O CH<sub>3</sub> Q.18 The IUPAC name of compound  $CH_3 - C = C - C - H$  is:  $HO - C = O CH_3$ (A) 2-Amino-3 chloro-2-methylpent-2-enoic acid (B) 3-Amino-4-chloro-2-methylpent-2-enoic acid (C) 4-Amino-3-chloro-2-methylpent-2-enoic acid (D) All of the above Q.19 The IUPAC name of compound  $CH_2 - C - OH$   $CH_2 - C - OH$ (A) 1,2,3-Tricarboxypropan-2-ol (B) 2-Hydroxy propane-1,2,3- tricarboxylic acid (C) 3-Hydroxy-3-carboxypentane-1,5-dioic acid (D) None

Q.20 The IUPAC name of 
$$CH_3$$
 is

(A) 3-Methyl cyclobut-1-ene-2-ol(C) 4-Methyl cyclobut-1-ene-3-ol

Q.21 The IUPAC name of compound

(A) 2-(Hydroxy methyl) methyl propanedioate

- (B) Methyl-2-(hydroxy methyl) propanedioate
- (C) 2-(Hydroxy methyl) dimethyl propanedioate
- (D) None of these
- Q.22 The suffix of the principal group, the prefixes for the other groups and the name of the parent in the structure

$$\begin{array}{c|c} \mathrm{HO-CH}_2-\mathrm{CH-CH}=\mathrm{C-CH}_2-\mathrm{C-C-OH}\\ & | & | \\ \mathrm{CH}_3 & \mathrm{Cl} & \mathrm{O} & \mathrm{O} \end{array}$$

(A) -oic acid, chloro, hydroxy, oxo, methyl, hept-4-ene
(B) -oic acid, chloro, hydroxy, methyl, oxo, hept-4-ene
(C) -one, carboxy, chloro. methyl, hydroxy, hept-4-ene
(D) -one, carboxy, chloro, methyl, hydroxy, hept-4-ene

- **Q.23** The IUPAC name of  $\beta$ -ethoxy- $\alpha$ -hydroxy propionic acid (trivial name) is:
  - (A) 1,2-Dihydroxy-1-oxo-3-ethoxy propane
  - (B) 1-Carboxy-2-ethoxy ethanol
  - (C) 3-Ethoxy-2-hydroxy propanoic acid
  - (D) All above

(B) 4-Methyl cyclobut-2-ene-1-ol(D) 2-Methyl cyclobut-3-ene-1-ol

## SECTION - III : ASSERTION AND REASON TYPE

Each question has 5 choices (A), (B), (C), (D) and (E) out of which ONLY ONE is correct.

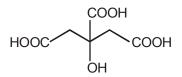
(A) Statement-1 is true, Statement-2 is true and Statement-2 is correct explanation for Statement-1.

(B) Statement-1 is true, Statement-2 is true and Statement-2 is not correct explanation for Statement-1.

(C) Statement-1 is true, Statement-2 is false.

(D) Statement-1 is false, Statement-2 is true.

- (E) Both Statements are false.
- Q.27 Statement-1: The IUPAC name of citric acid is 2-hydroxy propane 1,2,3,- tricarboxylic acid



**Statement-2**: When an unbranched carbon chain is directly linked to more than two like functional groups, then it is named as derivative of parent alkane which does not include the C-atoms of the functional groups.

- Q.28 Statement-1 : The IUPAC name for the compound, OHC–CH<sub>2</sub>–CH<sub>2</sub>–COOH is butane -3-formyl-1-oic acid Statement-2 : COOH is considered as substituent group while CHO is considered as the principal functional group.
- Q.29 Statement-1 : The IUPAC name for the compound  $C_6H_5$  COOCH<sub>2</sub>CH<sub>2</sub>COOH is 3-benzoyloxypropanoic acid. Statement-2 :  $C_6H_5$ CH<sub>2</sub>O is called benzoyloxy group
- Q.30 Statement-1 : Pentane and 2-methyl pentane are homologues Statement-2 : Pentane is straight chain alkane, while 2-methyl pentane is a branched chain alkane.
- Q.31 Statement-1 : Butane and 2-methyl butane are chain isomers Statement-2 : Butane is a straight chain alkane while 2-methyl butane is a branched chain alkane.
- Q.32 Statement-1 : Neopentane is chain isomer of n-pentane. Statement-2 : Molecular formula of neopentane and n-pentane is  $C_{s}H_{1,s}$ .

## **SECTION - IV : COMPREHENSION TYPE**

#### Comprehension

A saturated hydrocarbon (P) has six membered ring. Three alkyl groups attached to the ring alternate to each other.

(i) First group has only two carbon atoms.

(ii) Second group has four carbon atoms and its all hydrogen atoms are chemically same.

(iii) Third group has total five carbon atoms. Its main chain contains three carbon atoms with ethyl as a substituent.

- Q.33 How many 3° hydrogen atoms are present in the hydrocarbon (P) ?
  (A) 2
  (B) 3
  (C) 4
  (D) 5
  Q.34 How many 2° carbon present in the compound (P).
  - (A) 10 (B) 12 (C) 6 (D) 8
- Q.35 IUPAC name of hydrocarbon (P) is

(A) 1–(1–Ethylpropyl)–3–ethyl–5–(1,1–dimethylethyl)cyclohexane

**(B)** 1–Ethyl–3–(1–ethylpropyl)–5–(1,1–dimethylethyl)cyclohexane.

(C) 1–(1,1–Dimethylethyl)–3–ethyl–5–(1–ethylpropyl)cyclohexane

(D) 1-(1,1-Dimethylethyl)-3-ethyl-5-(2-ethylpropyl)cyclohexane

# **ANSWER KEY**

#### EXERCISE - 1

 1. A
 2. C
 3. C
 4. B
 5. C
 6. D
 7. B
 8. C
 9. D
 10. B
 11. A
 12. B
 13. B

 14. A
 15. A
 16. B
 17. D
 18. C
 19. B
 20. D
 21. B
 22. B
 23. A
 24. D
 25. B
 26. D

 27. C
 28. C
 29. B
 30. A
 31. C
 32. C
 33. D
 34. B
 35. C
 36. C
 37. D
 38. B
 39. C

 40. D
 41. B
 42. A
 43. A
 44. B
 45. D
 46. B
 47. C
 48. B
 49. D
 50. B
 51. C
 52. B

 53. B
 54. C
 55. C
 56. A

#### EXERCISE - 2 : PART # I

<b>1.</b> A, B, C	<b>2.</b> A, B, C	<b>3.</b> A, B, C, D	<b>4.</b> A, B, C, D	<b>5.</b> A, B, C, D	<b>6.</b> A, B, C	<b>7.</b> C, D
<b>8.</b> A, B, C, D	<b>9.</b> A, B, C, D	<b>10.</b> A, B, D	<b>11.</b> A, B	<b>12.</b> B,C	<b>13.</b> A, B, C	<b>14.</b> A, B, C
<b>15.</b> A, B, C	16. B,D	17. A, C				

#### PART # II

1. A 2. B 3. B 4. A 5. A

#### EXERCISE - 3 : PART # I

1.  $A \rightarrow (s), B \rightarrow (q, r), C \rightarrow (p), D \rightarrow (q, r)$ 2.  $A \rightarrow (r), B \rightarrow (s), C \rightarrow (p), D \rightarrow (q)$ 3.  $A \rightarrow (q), B \rightarrow (r), C \rightarrow (s), D \rightarrow (p)$ 4.  $A \rightarrow (r), B \rightarrow (s), C \rightarrow (p), D \rightarrow (q), E \rightarrow (u), F \rightarrow (t)$ 5.  $A \rightarrow (r, q), B \rightarrow (p), C \rightarrow (s)$ 6.  $A \rightarrow (r), B \rightarrow (p), C \rightarrow (s), D \rightarrow (q)$ 7.  $A \rightarrow (q, r), B \rightarrow (r, s), C \rightarrow (p)$ 8.  $A \rightarrow (q), B \rightarrow (r), C \rightarrow (s), D \rightarrow (p)$ 

#### PART # II

 Comprehension #1:
 1.
 T
 2.
 F
 3.
 F
 4.
 T
 5.
 T

 Comprehension #2:
 1.
 C
 2.
 B
 3.
 B
 4.
 A

#### **EXERCISE - 5 : PART # I**

**1.** 3 **2.** 3 **3.** 2 **4.** 4 **5.** 1 **6.** 4 **7.** 1 **8.** 4 **9.** 1

### PART # II

1. 3-Aminobenzoic acid		2.	В	<b>3.</b> 4-Methylbenzensulphonic acid <b>4.</b>			4.	С	5.	В	
6.	5	7.	В	8.	D	9.	B,C				

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